

Effects of Nanomaterials on Engineering Performance of a Silicate-Based Sealer for Cementitious Composite

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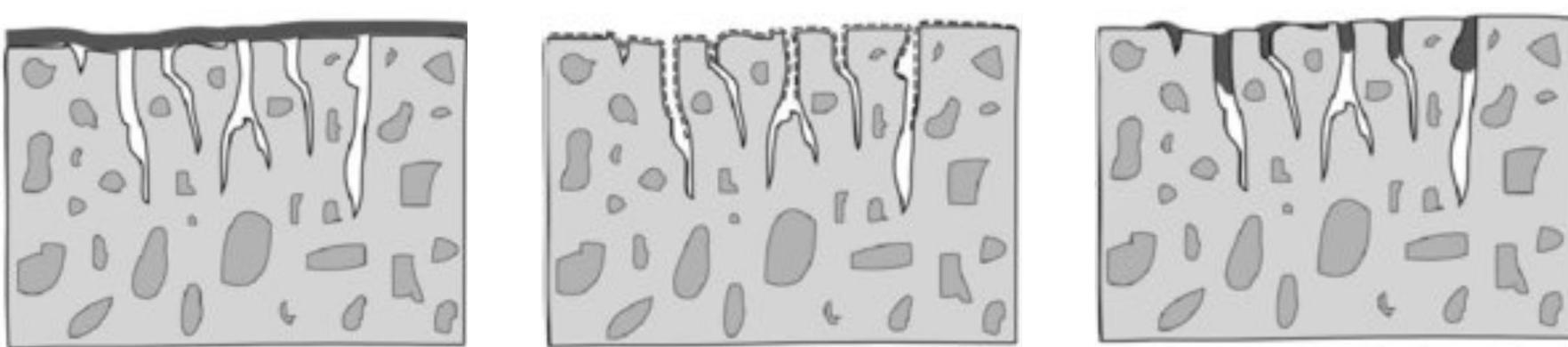
Surface treatments

Effectively protect concrete from salt scaling and extend the service life of concrete pavements, bridge decks, ...

Barrier coatings: *epoxy, acrylics, ...*

Hydrophobic impregnation: *silane, siloxane, ... (3 to 4 years)*

Pore blocker: *soy methyl ester, sodium silicate, ...*



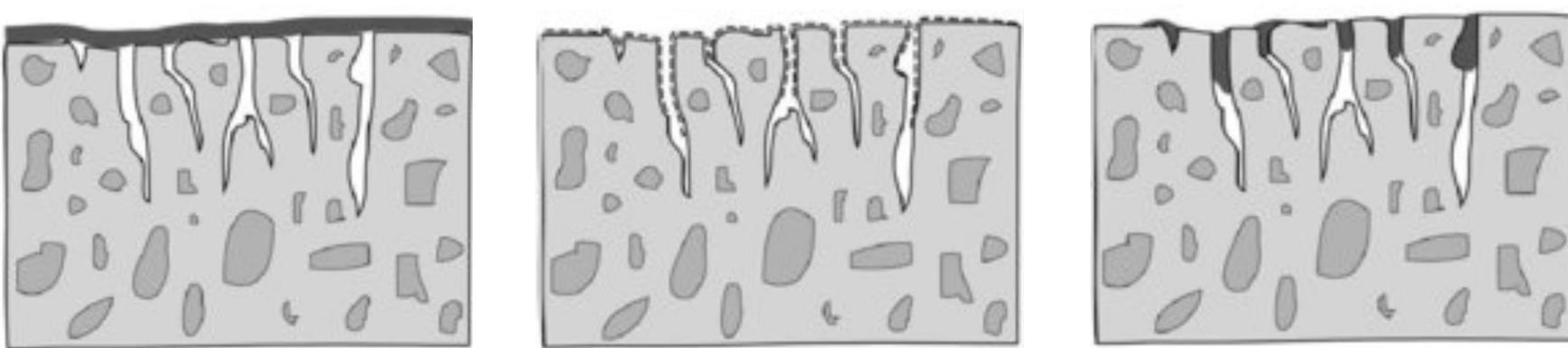
(Diamond S, Berke NS. 1997)

Introduction

Potassium methyl siliconate

as surficial hydrophobic treatment of wood, cellulose, etc.

- ➔ Reacts with airborne CO₂ and generates a compact, cross-linked, and insoluble methyl-silicone membrane
- ➔ Provides the substrate with a hydrophobic and denser surface
- ➔ Serves as a pore blocker: methyl-silicone membrane, potassium carbonate or bicarbonate



(Diamond S, Berke NS. 1997)

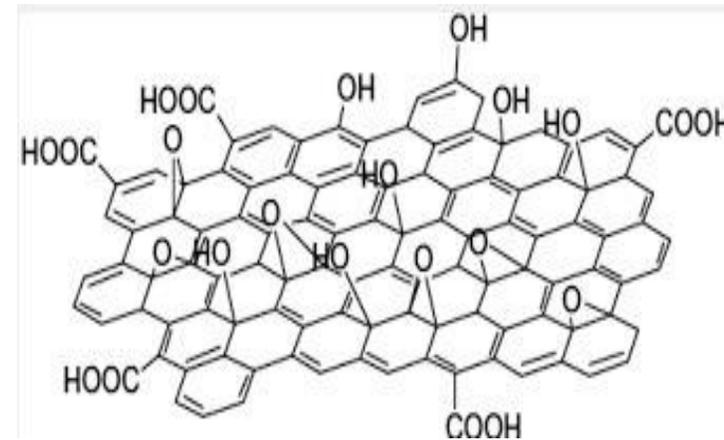
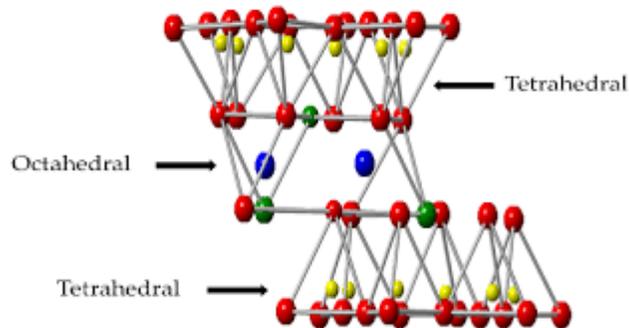
THE WORLD'S GATHERING PLACE FOR ADVANCING CONCRETE

Nano-enhanced Sealer

as surficial hydrophobic treatment of wood, cellulose, etc.

- ➔ Nano-SiO₂/siloxane sealer/mortar: water absorption ↓~20%, gas permeability ↓~28%
(Li et al. 2018)
- ➔ Nanoclay/silane sealer/concrete: water permeability ↓~39%
(Woo et al. 2008)

Graphene oxide & montmorillonite nanoclay



Hypothesis

The admixed GO and montmorillonite NC can provide the PMS sealer w/ better performance:

- ➔ GO/NC modify the PMS sealer and improve its hydrophobicity: via nano/micro-roughness.
- ➔ NC may react with alkaline PMS sealer to generate K-A-S-H gel to refine the microstructure of mortar matrix.
- ➔ GO may catalyze the process in hypothesis 2.

Raw materials

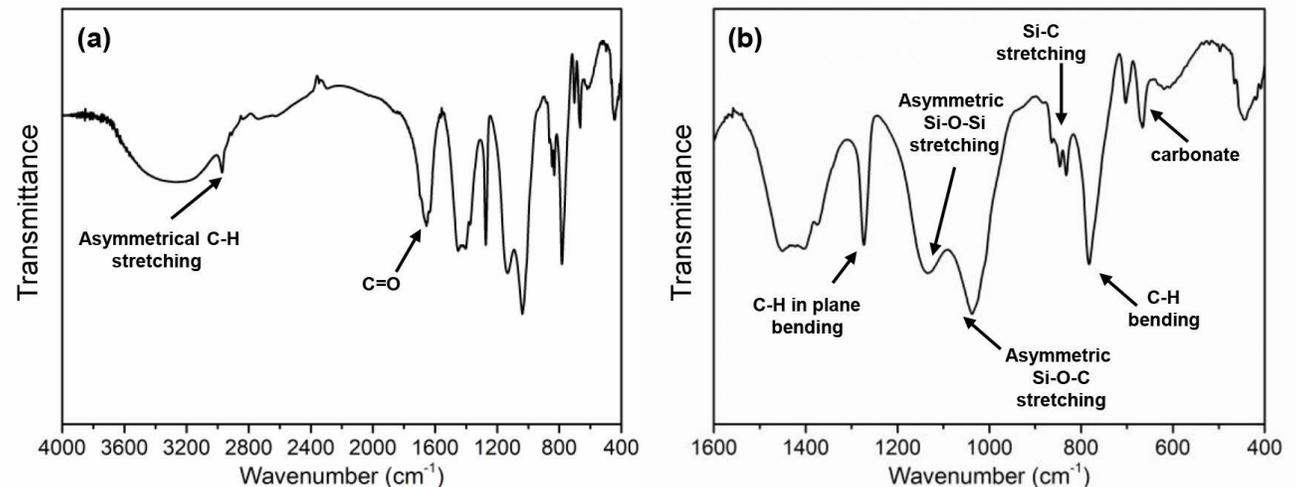
NC

- bulk density: 0.678 g/cm^3
- aspect ratio: 200 to 400
- Lateral size: ~ 20 to $25 \text{ }\mu\text{m}$
- Si-O tetrahedron sandwiched by 2 layers of Al-O octahedron

GO: 71 wt.% C + 26 wt.% O

- lateral size: 4 to $20 \text{ }\mu\text{m}$
- zeta potential: -30 mV at pH 7.0

A penetrating PMS sealer:
PMS/siloxane resin hybrid



Nano-sealer



Branson digital sonifier
(S-450D, 400W, and 60% amplitude)

To obtain a well-dispersed nano-modified PMS sealer suspension

- 30 min ultrasonic dispersion
- GO: 0.015%, 0.03%, **0.06%**, 0.1%, 0.15%
- Based on water contact angle of GO-modified PMS sealer then introduce NC.

1.5N-6G-P, 3N-6G-P, 6N-6G-P, 10N-6G-P, **15N-6G-P**

- NC: 0.015%, 0.03%, 0.06%, 0.1%, 0.15% by the mass of PMS sealer

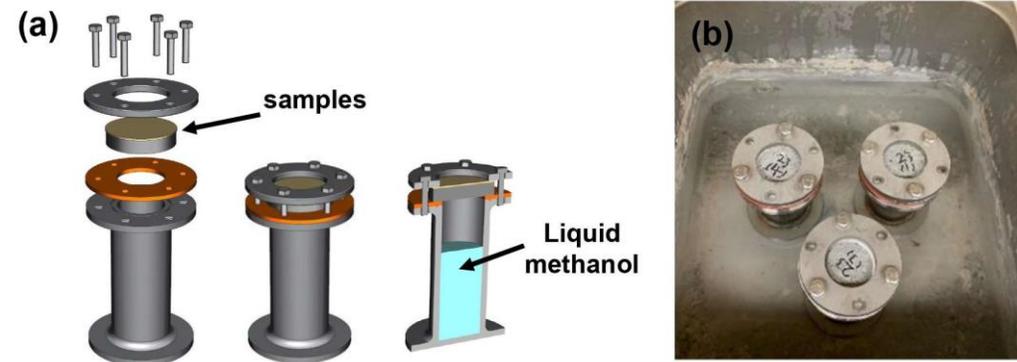
15N-6G-P: 0.15% NC+0.06%GO+PMS



Experimental test

- **Water absorption test:** 5 cm (D) × 2 cm (H) (ASTM C1585-20)

- **Gas permeability test:**



- **Water contact angle (ASTM D7334)**

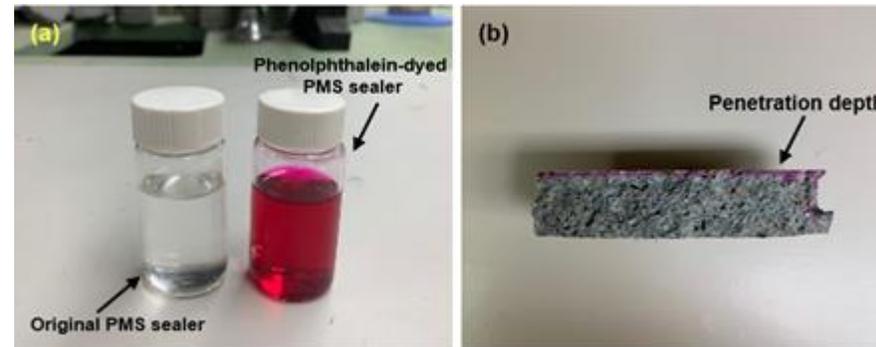
- **Penetration depth vs. Viscosity**

- **Salt scaling test:**

F/T+W/D cycles in 3.5 wt% NaCl

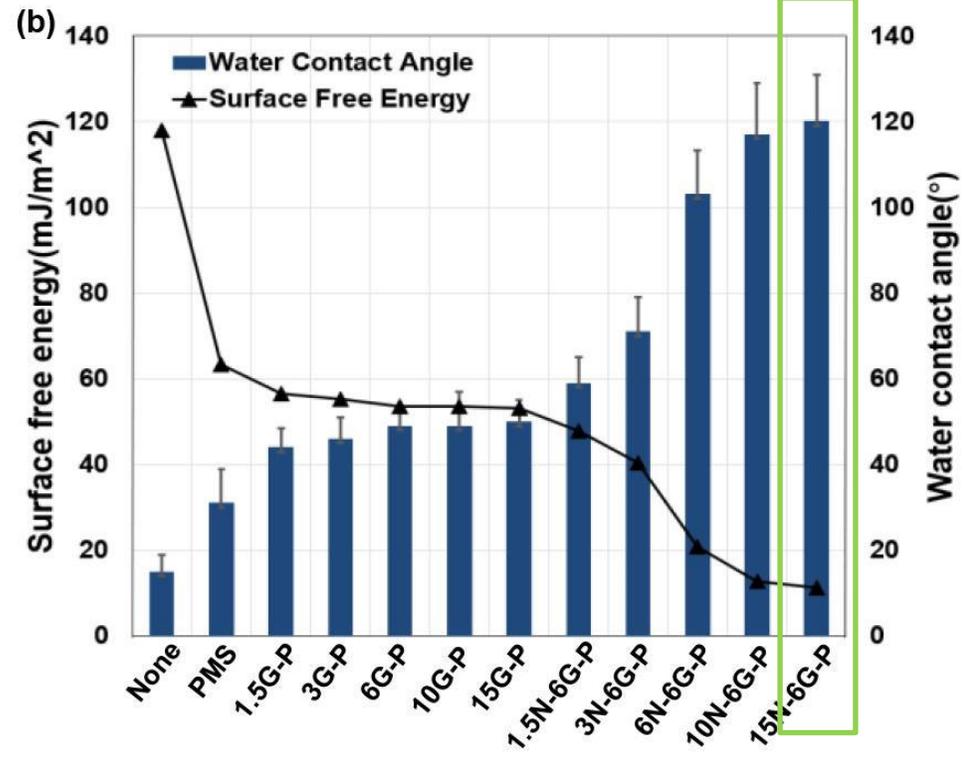
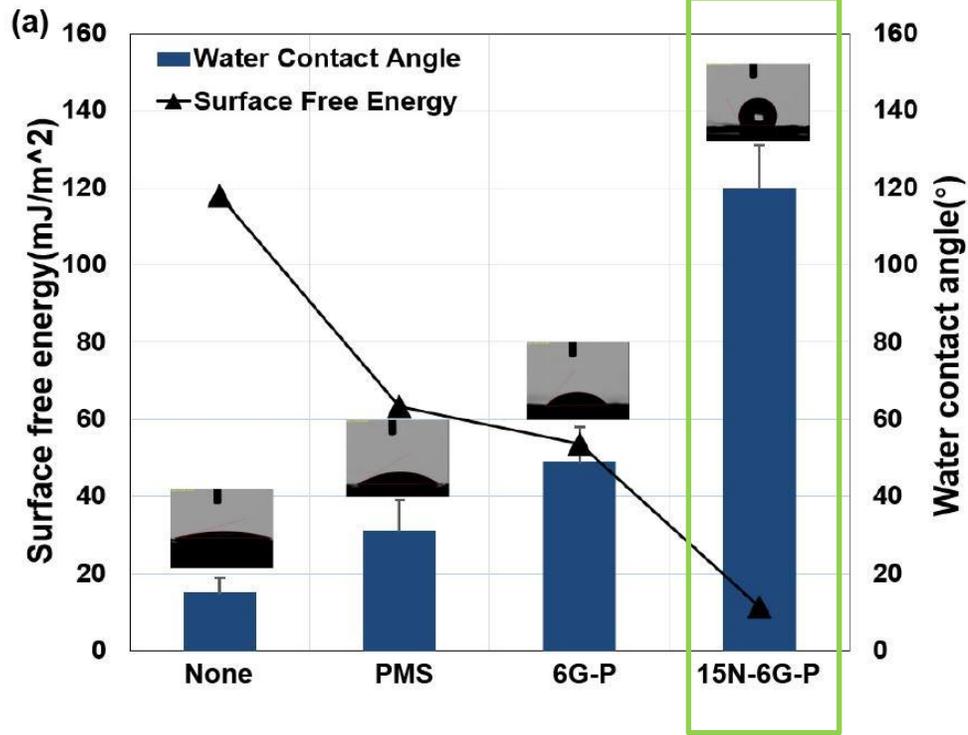
- **High-resolution optical microscope**

- **TGA/DTG:** 50°C to 500°C, 10°C/min



Results & Discussion

Water contact angle and surface free energy

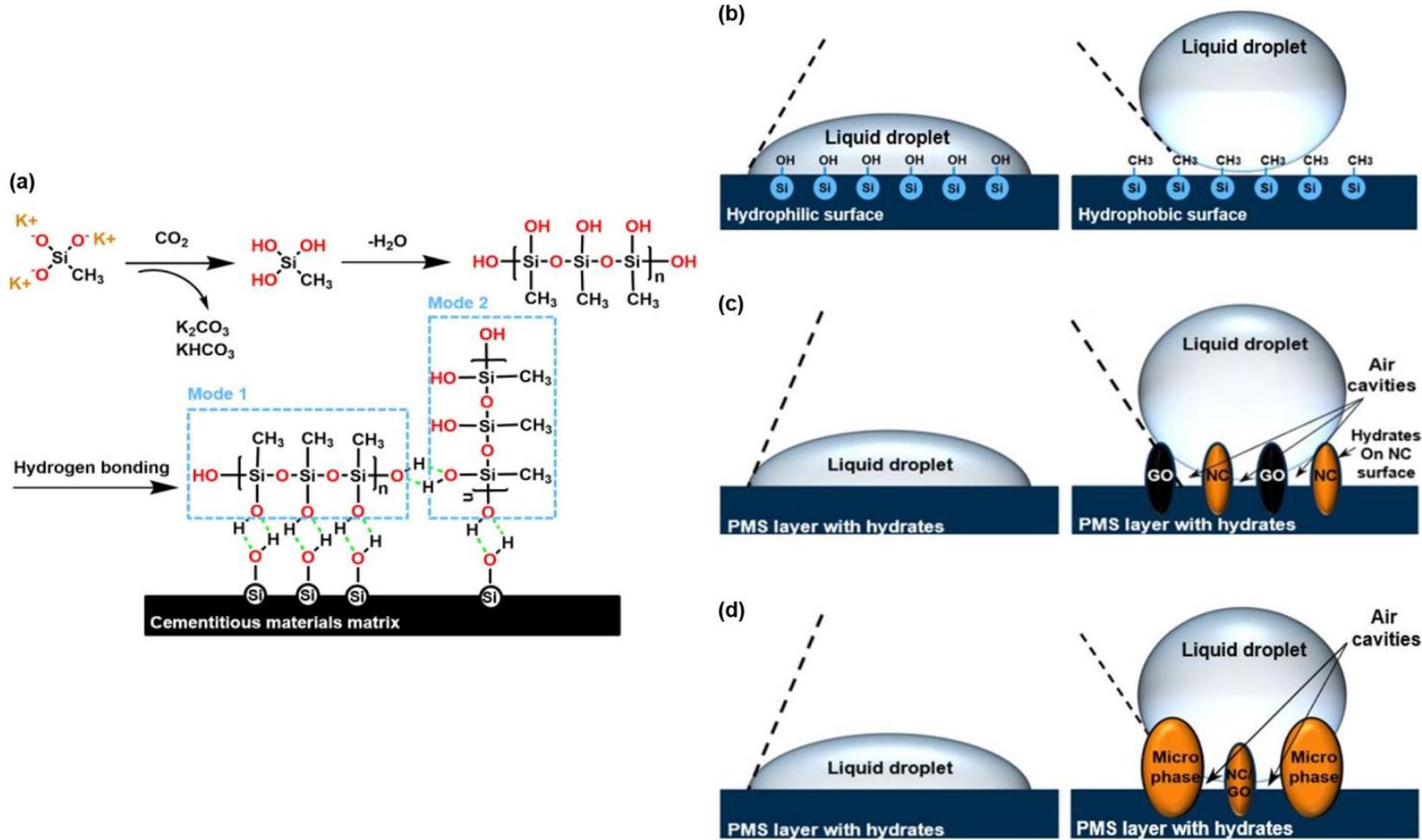


GO and NC reduced the hydrophilicity and decreased the surface free energy of PMS sealer membranes

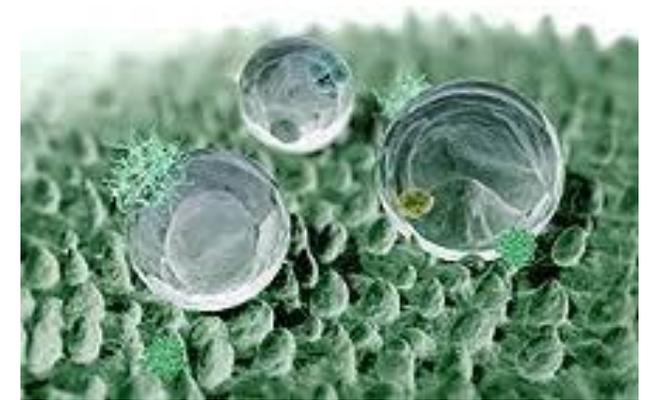


Results & Discussion

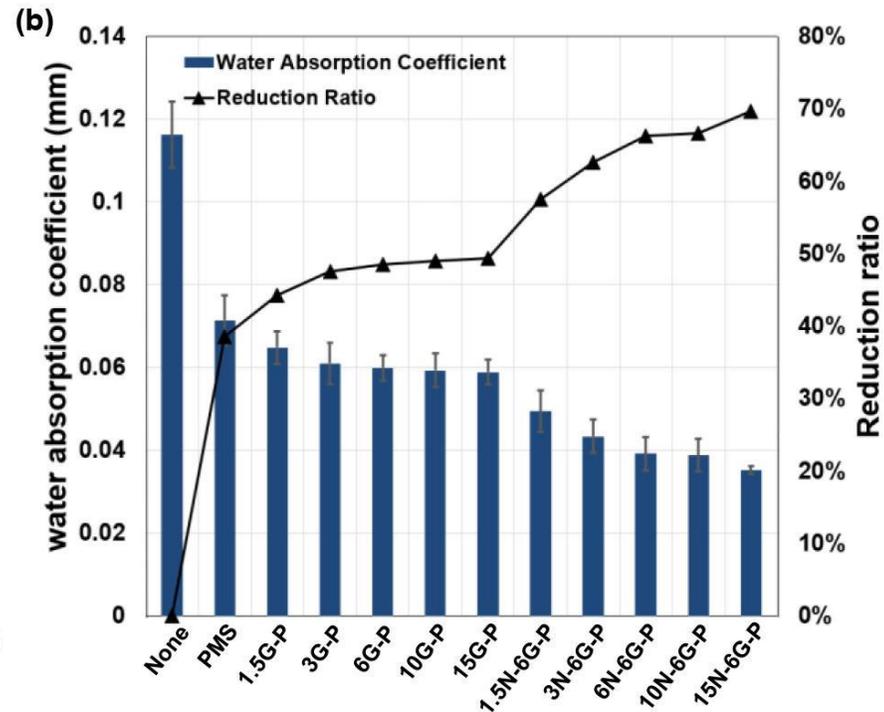
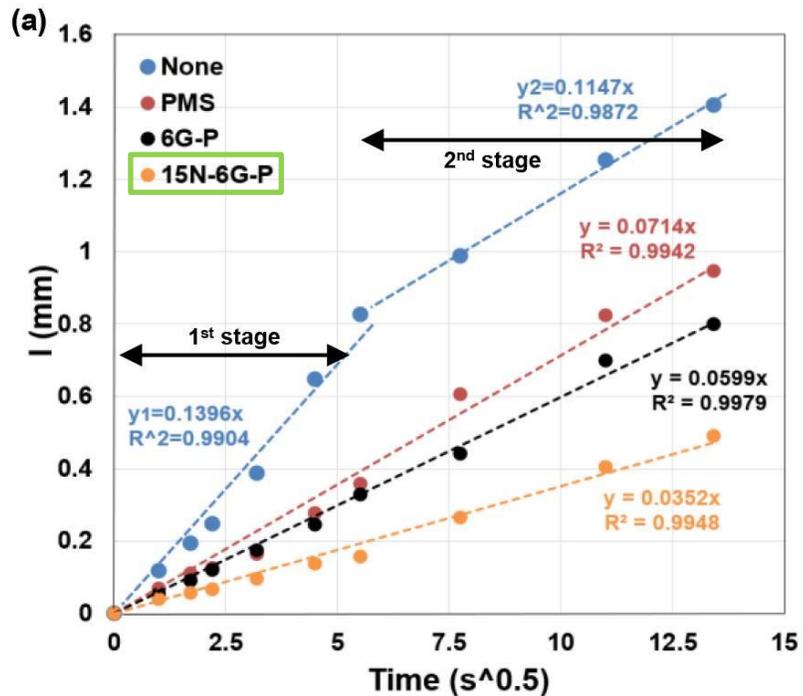
Water contact angle and surface free energy



a rough nano-/micro-scale hierarchical structure due to agglomeration of nanomaterials



Water absorption behavior

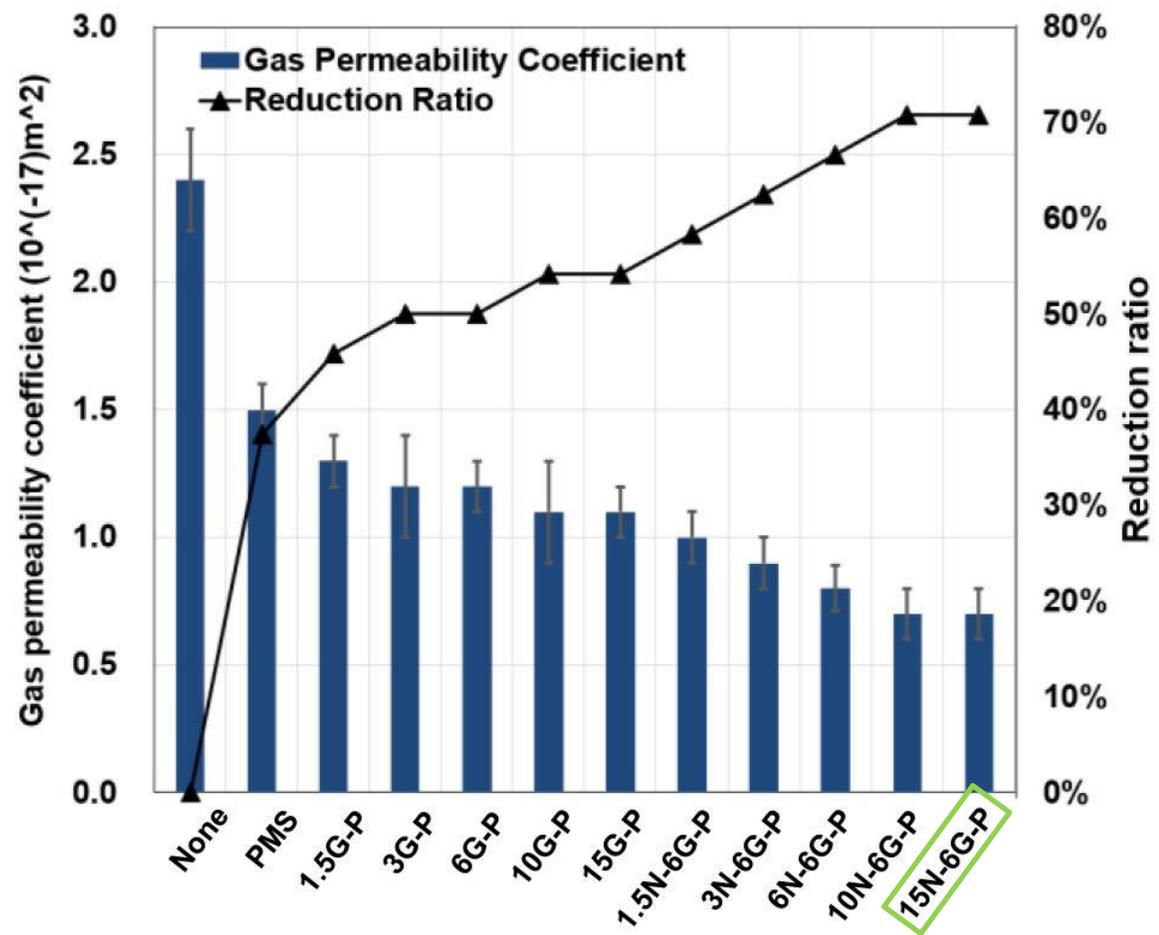


The application of original PMS-based sealer, 6G-P, and 15N-6G-P decreased the k of mortar surface by 39%, 48%, and **70%**, resp.

- ✓ The denser microstructure of mortar substrate induced by NC/GO
- ✓ The increased hydrophobicity of nano-modified sealer treated surface

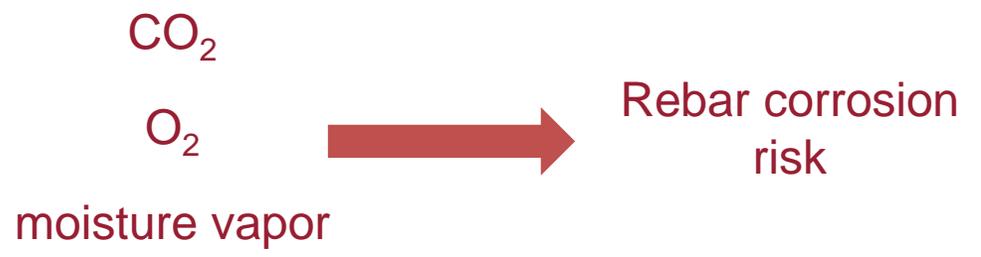
Results & Discussion

Gas permeability

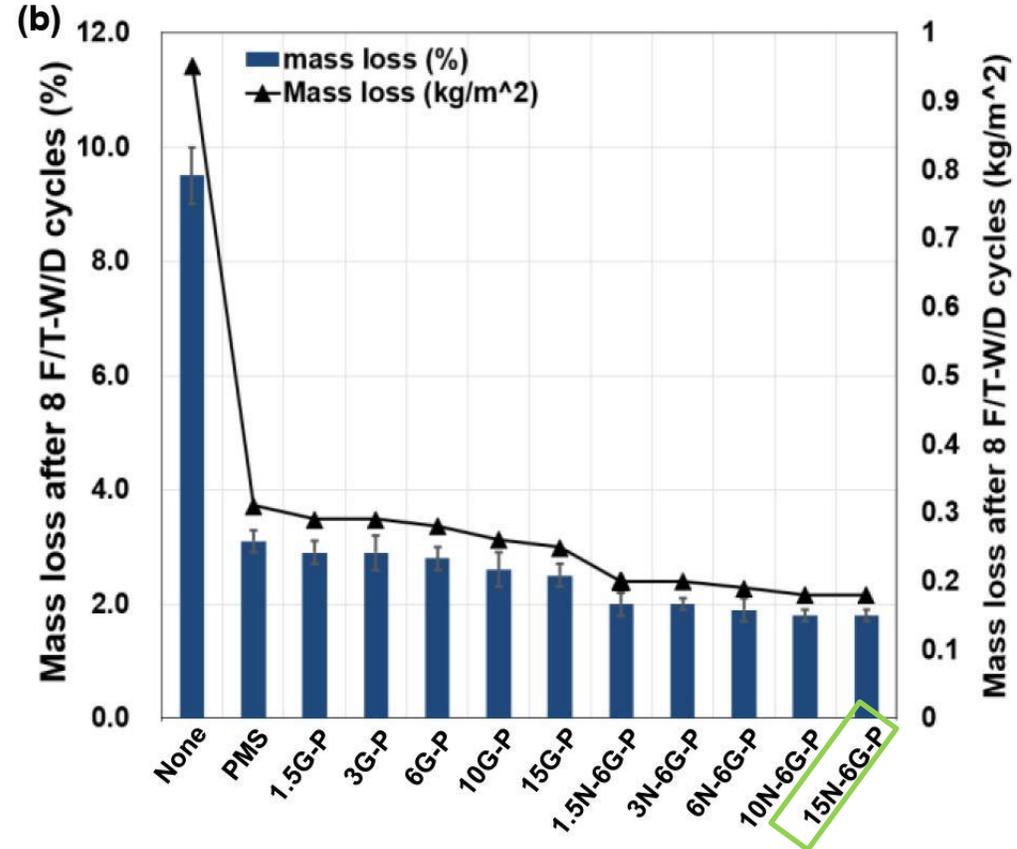
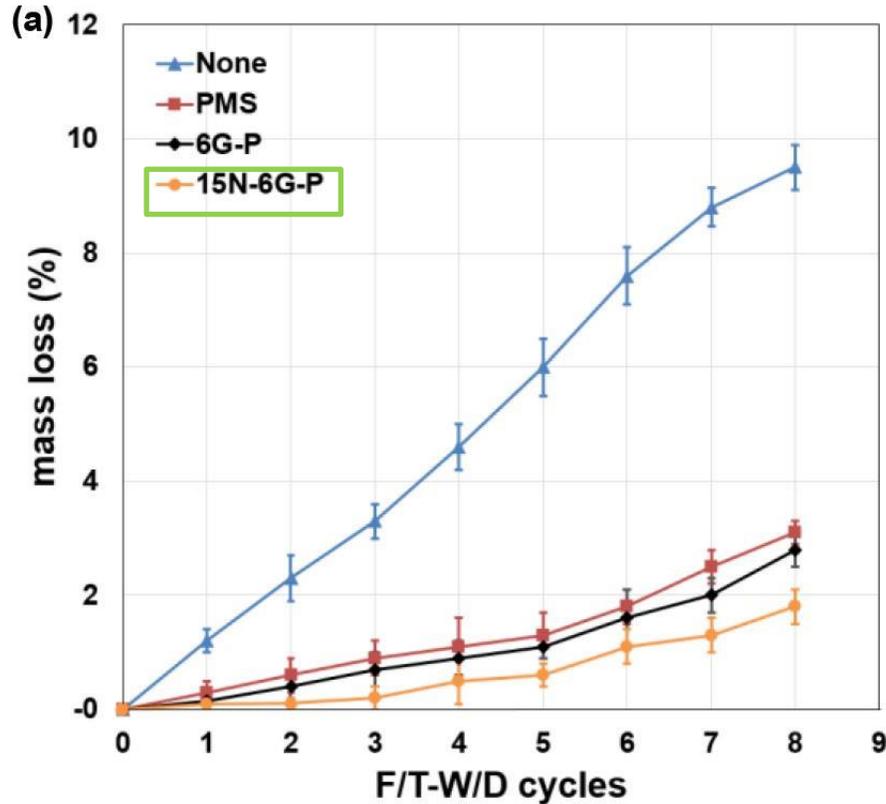


The application of original PMS-based sealer, 6G-P, and 15N-6G-P decreased the *G* of mortar surface by 38%, 54%, and **71%**, resp.

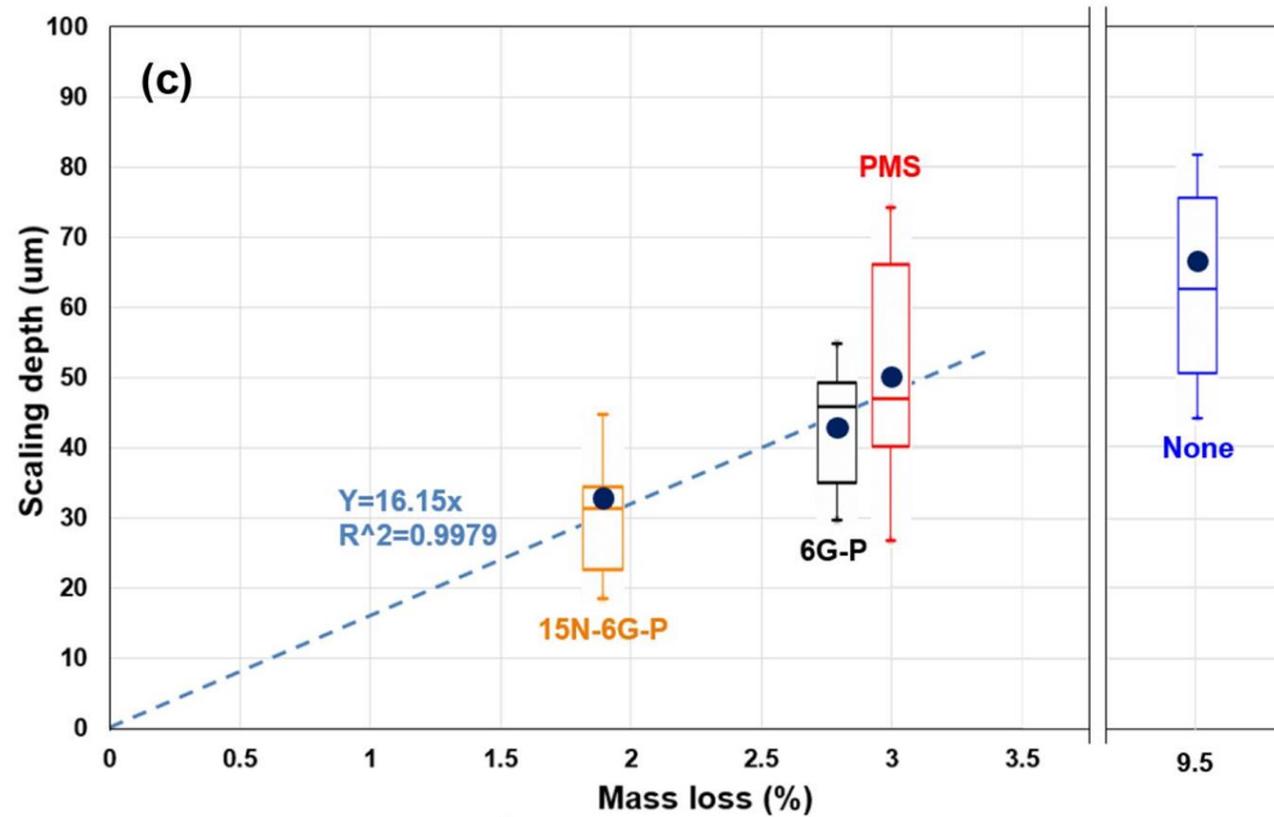
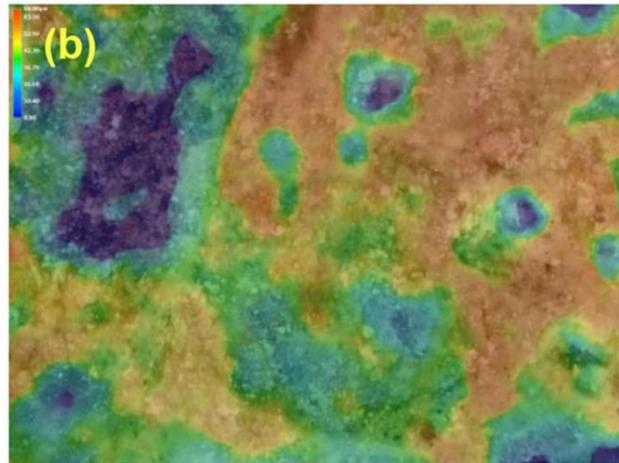
✓ The denser microstructure of mortar substrate induced by NC/GO



Salt-scaling resistance

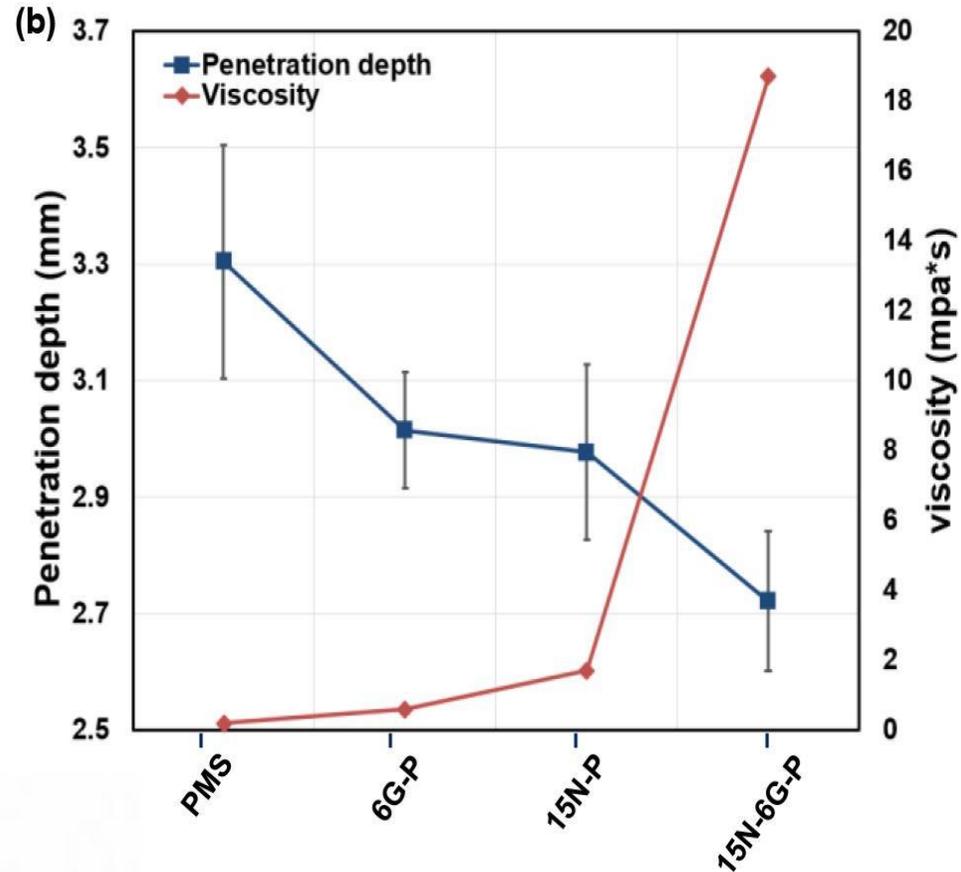
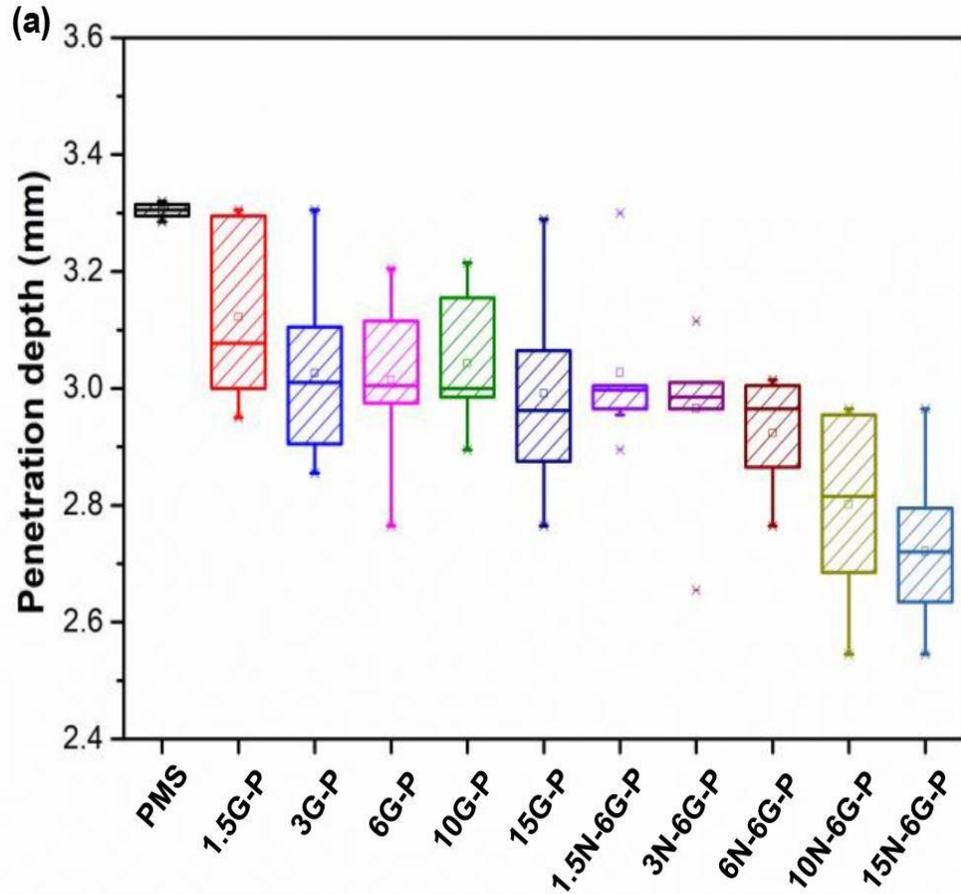


Salt-scaling resistance



Results & Discussion

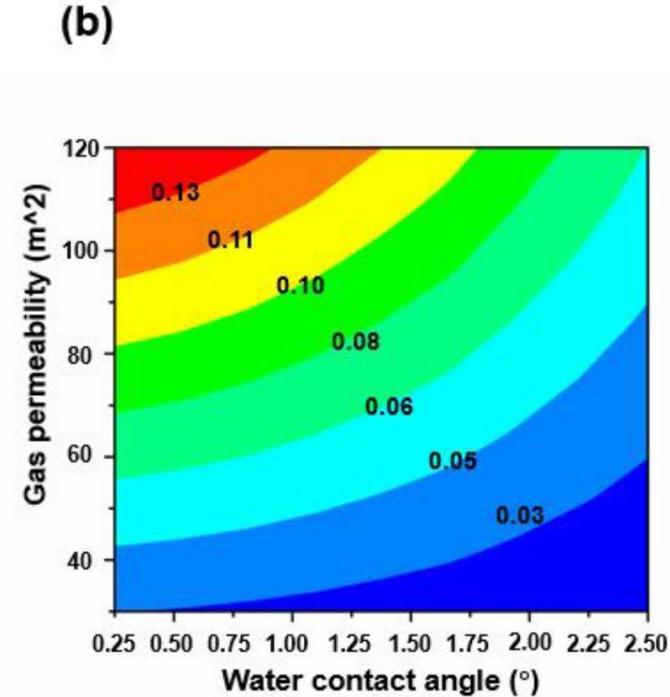
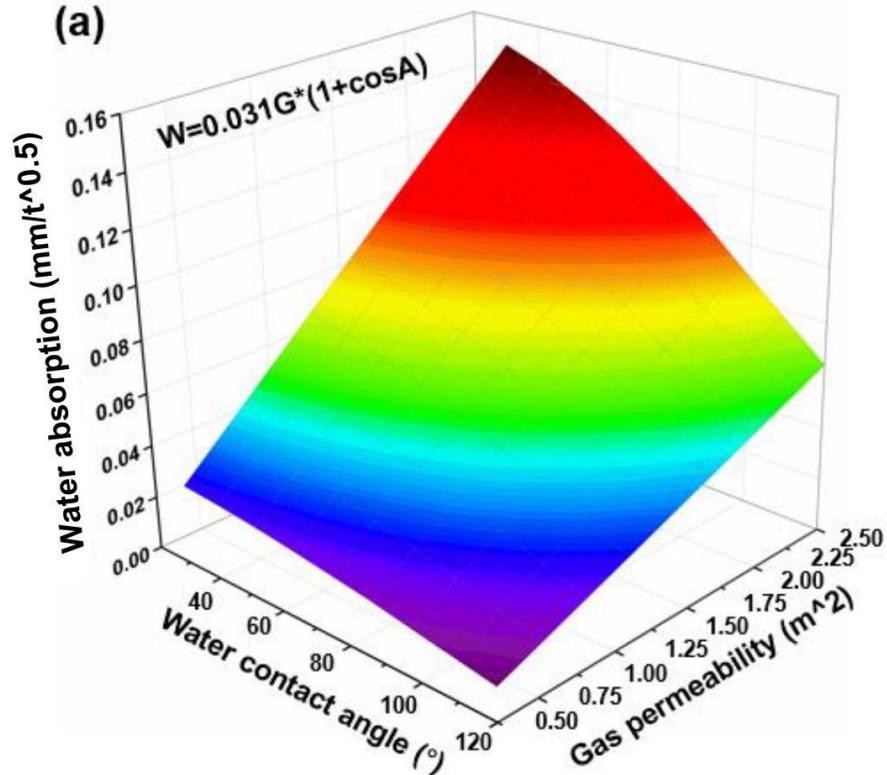
Penetration depth vs. viscosity



sealer application rate of 0.136 L/m²

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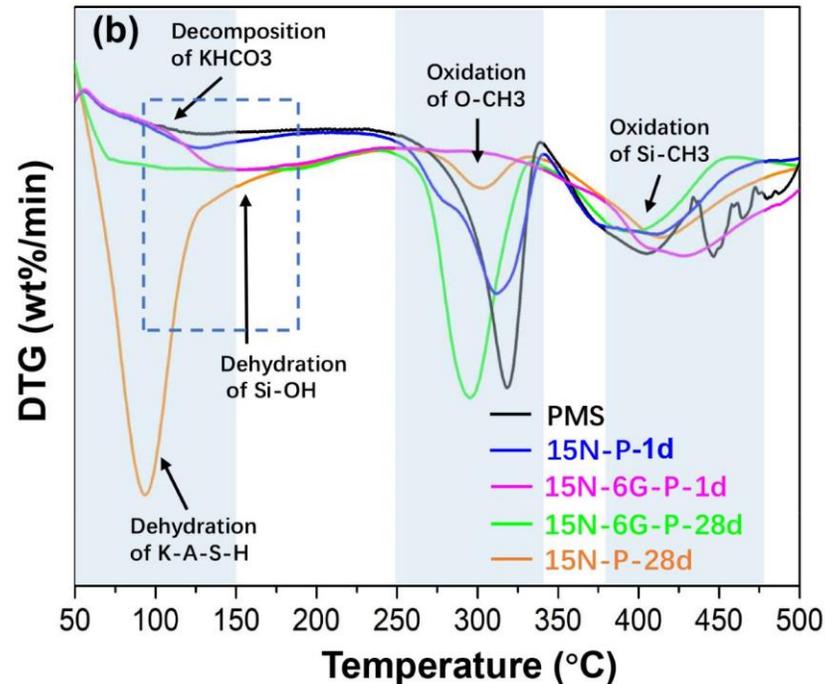
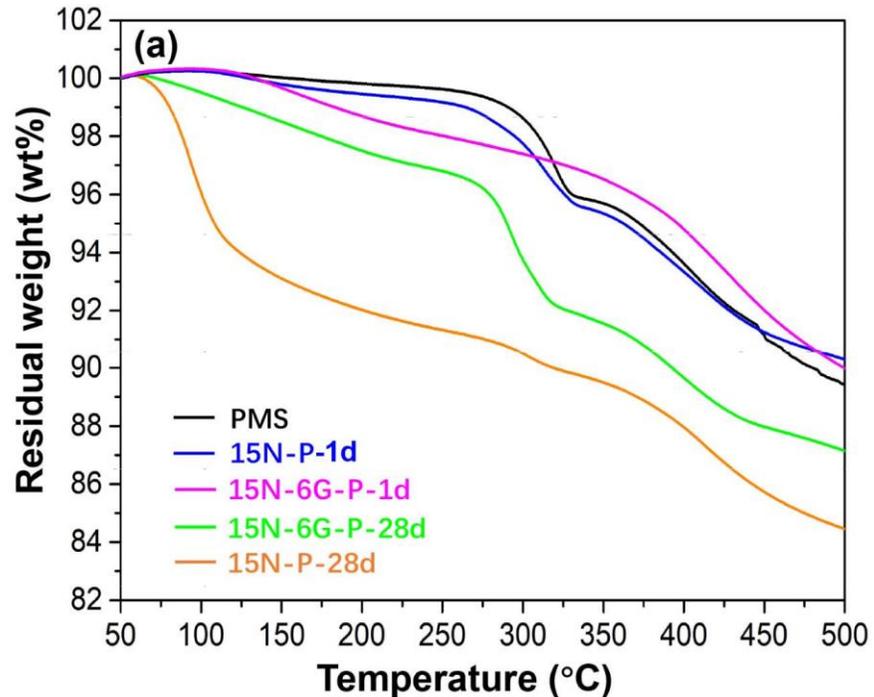
Relationship between parameters



A decreased gas permeability & an increased water contact angle, together, would correspond to a lower water absorption coefficient.

A more diverse and comprehensive experimental dataset is needed to develop more reliable and more accurate models.

Thermogravimetric analyses of various sealers



250°C - 340°C: volatile CH₃OH from the oxidation of -O-CH₃

370°C - 470°C: CO₂ from the oxidation of Si-CH₃

15N-6G-P-1d: The admixed GO prevented the oxidation of O-CH₃, likely due to the strong hydrogen bonding btwn the hydrophilic GO and -O-CH₃

The PMS sealer preferentially reacted with the admixed NC to produce K-A-S-H gel, ↓ ↓ its availability to react with CO₂ to produce KHCO₃.

Conclusion & summary

This study demonstrated synergistic benefits of modification by nanoclay and GO to the PMS-based sealer

- ✓ The mortar coated with the **0.15wt% NC+0.06wt.% GO+PMS hybrid** (15N-6G-P) sealer represented the best performances, featuring its water contact angle of 120° (vs. 32°)
- ✓ A **reduction of ~70%** in both the water absorption coefficient and gas permeability coefficient of the mortar, vs. ~38% induced by the original sealer
- ✓ After 8 F/T+W/D cycles in 3.5 wt.% NaCl, the 15N-6G-P sealed mortar saw a **1.9% mass loss** (vs. 3.0% for the original PMS sealed mortar and 9.5% for the unsealed mortar) and greatly reduced scaling depth

Conclusion & summary

- ✓ **Hydrophobicity**: -OH replacement by -CH₃; and micro-/nano-roughness induced by the admixed nanomaterials
- ✓ **Refined the pores (film /pore blocker)** of the mortar; NC reacted with the alkaline PMS sealer to produce more hydrates (K-A-S-H gel)
- ✓ A decreased gas permeability and an increased water contact angle, together, would correspond to a **lower water absorption coefficient**
- ✓ The lower the water absorption coefficient, the **better resistance to salt scaling**

Future work: concrete; long-term performance; optimization; life-cycle assessment in typical service environments

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TriDurLE

National Center for Transportation
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Questions?

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