

Modeling Hydration Kinetics of Sustainable Cementitious Binders Using a Data Informed Nucleation and Growth Approach

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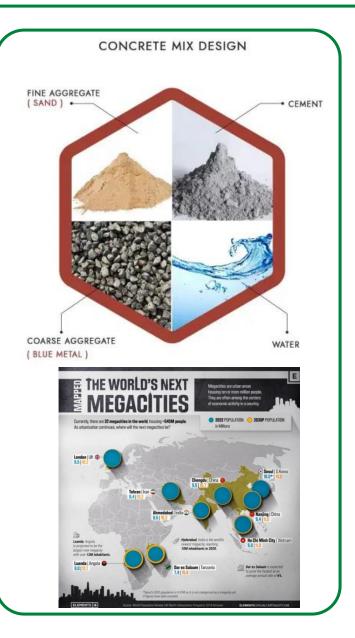






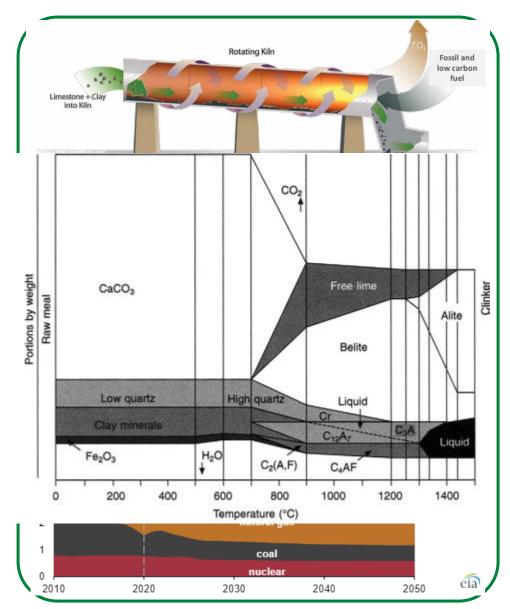
Concrete: Overview

- Concrete: The principal material for construction of (nearly all) infrastructure
- Concrete = Portland cement + water + sand
 + stones + chemicals (to regulate properties)
- Production-and-use: 40 billion tons/year
 - Employs 10 million Americans
 - Creates \$1.3 trillion worth of engineered systems
- 2050 projections
 - Global population: 10 billion
 - (Sub)urbanites: 6.6 billion
 - Rise of >50 megacities
 - Concrete production: ~60 billion tons/year





- The carbon-footprint problem
- Limestone (CaCO₃) is needed to produce cement: 70%_{mass}
 - CO_2 released at ~800 °C
- Clinkering temperature: 1450°C
 - Achieved using fossil fuels
 - Switching to electricity doesn't help (68% of electricity is generated from fossil fuels)
- 0.85 tons of CO₂ emitted for every ton of cement produced
- Cement production: 8% of all anthropogenic CO₂ emissions
- IPCC: "Essential field of action"

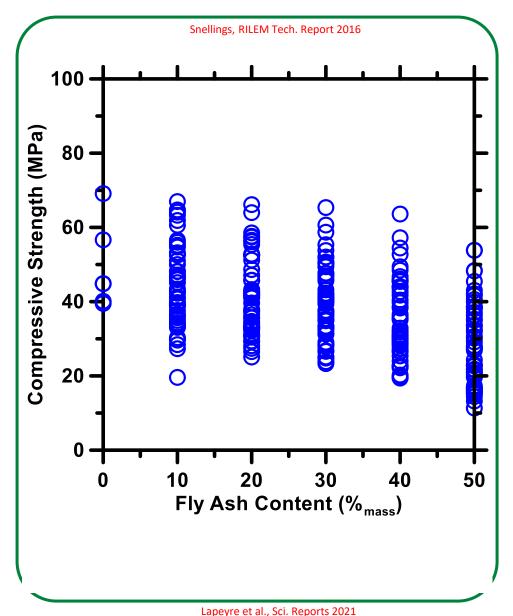


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Using supplementary cementitious materials

- Use supplementary cementitious materials (SCMs) to partially replace cement in concrete
 - Coal fly ash; slags; waste glass
 - Geological materials (e.g., clay)
- SCMs are not as reactive as cement: Cannot replace >50%
- Feature substantial batch-to-batch variations in composition
- Affect chemical reactions (cement hydration); microstructural evolution; property development in unpredictable ways.



Cook et al. Mat. & Des. 2021

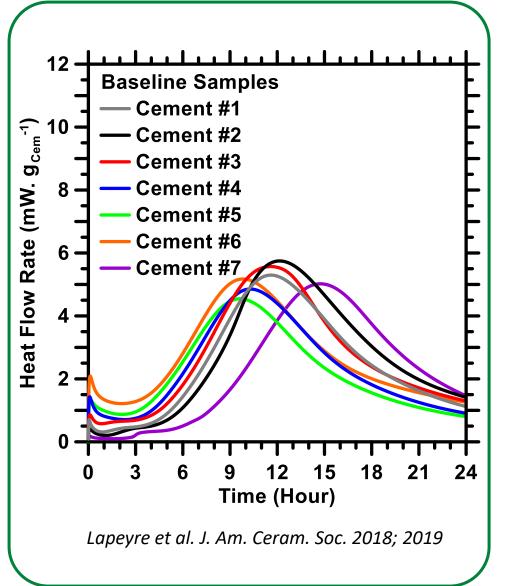
Han et al. Front. Mat. 2022

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Cement Hydration: Plain Paste

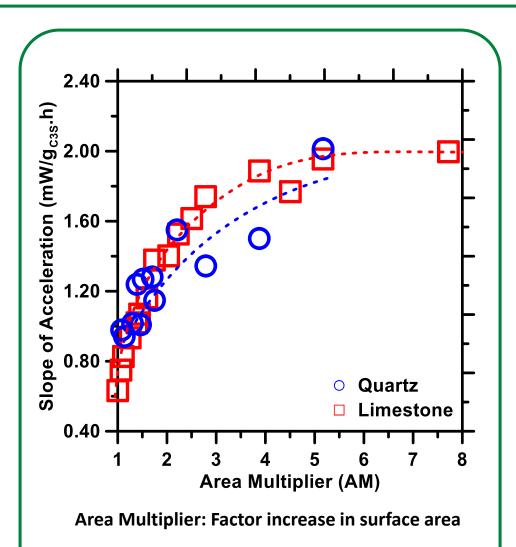
- Cement reacts with water to form products: "hydration"
- **Isothermal calorimetry** is typically used to monitor hydration kinetics
 - Heat evolution linked to rate/extent of cement hydration
- Anhydrous phases dissolve; products (C-S-H) nucleate and grow; microstructure evolves; and properties develop
- Different phases; disparate kinetics; disparate mechanisms
- Modeling challenges:
 - Account for hydration mechanisms of all anhydrous phases





Cement Hydration: Binary Paste

- SCMs: fillers (limestone and quartz)
- In general, filler addition increases reaction rates
- Extent of acceleration is strongly dependent on:
 - Filler fineness (surface area)
 - Cement replacement (%)
- Limestone *superior* than quartz at equivalent surface area
- Modeling challenges:
 - Not all fillers are the same
 - Acceleration does NOT linearly increase with surface area

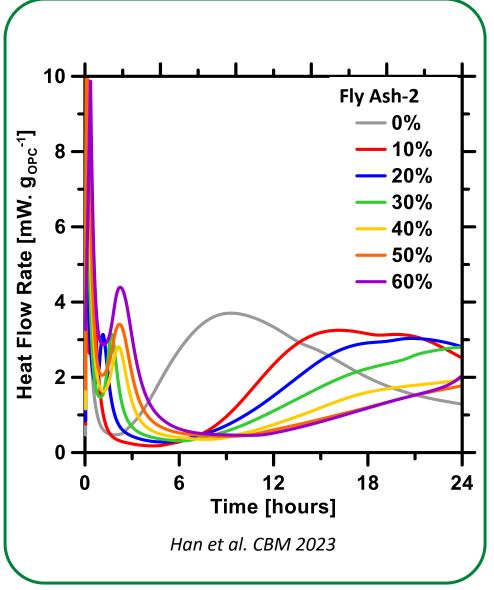


Oey et al.; Kumar et al. J. Am. Ceram. Soc. 2013; 2017



Cement Hydration: Binary Paste

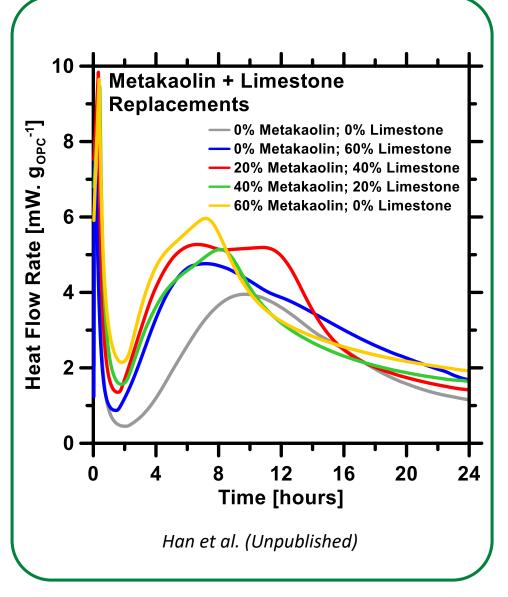
- SCMs: silica or aluminosilicates
- Silica fume and metakaolin at least in the first 24 h – accelerate cement hydration kinetics
 - Later on, they may react with calcium hydroxide; alter hydration
- Fly ash a highly heterogeneous material – alters hydration kinetics in myriad ways
- Modeling challenges:
 - Changes in hydration kinetics strongly dependent on PSD and chemistry of the SCM
 - Multiple mechanisms (filler effect; pozzolanic reaction; etc.) at play at any given time





Cement Hydration: Ternary Paste

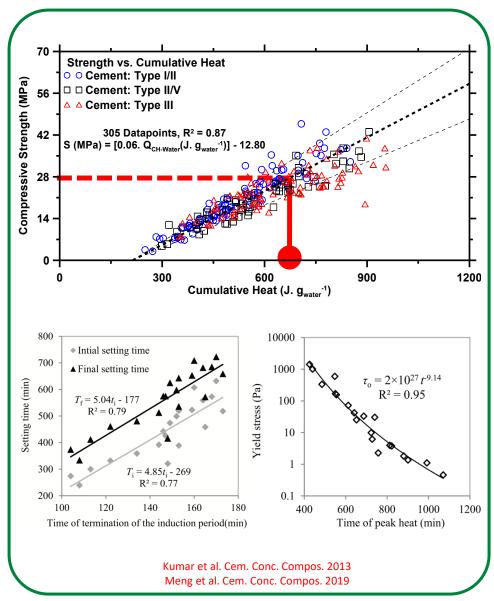
- Interactions between SCMs and paste components are complex
 - Filler effect; pozzolanic effect; secondary reactions; etc.
- E.g., Cement-limestone-metakaolin reaction to form hemi- and mon carboaluminate
- E.g., SO₃/Al₂O₃ of the binder could be altered, thereby affecting hydration of the C₃A phase
- Modeling challenges:
 - Multiple concurrent reaction mechanisms need to be considered to model the overall progression of reaction





Why model hydration kinetics?

- Apriori prediction of hydration kinetics (calorimetry profile) is very useful
- If we know what % of cement has reacted, thermodynamic models can predict phase assemblage
- More heat = greater extent of cement hydration = more products
 = less porosity; more solid-to-solid phase connectivity; more strength
- Cumulative heat is linked to properties
 - strength; set time; rheology; etc.
- If hydration can be predicted, performance can be estimated
- Useful for cement design, mixture proportioning, etc.

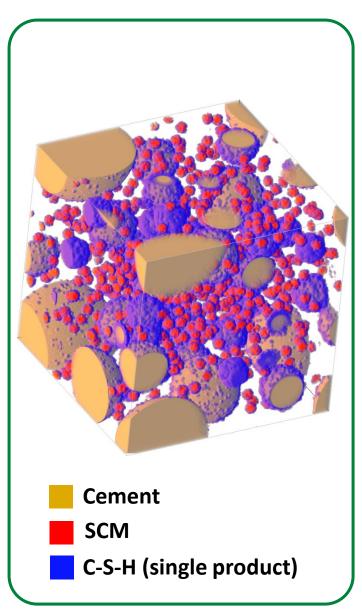


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Nucleation-and-Growth Mechanism

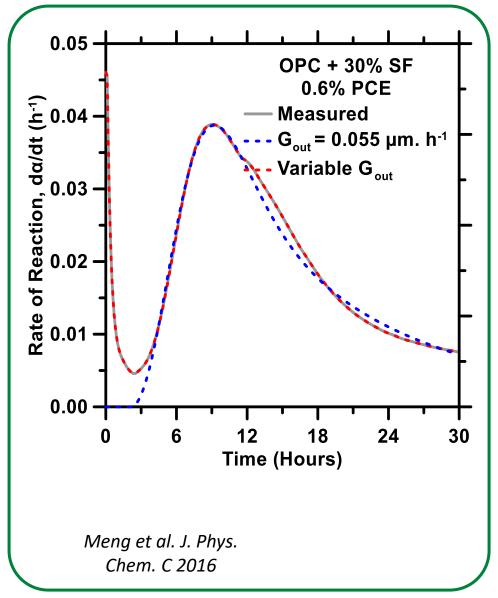
- Classical phase boundary (heterogenous) nucleation-and-growth models
 - substrate: cement and/or SCM surfaces
 - A single product with constant density forms heterogeneously at nucleation event
 - Product grows (usually in isotropic fashion) at constant rate
- Inputs
 - cement, SCM, water contents; SSA of solids
 - cement/SCM chemistry not considered
- Parameters need to be optimized using calorimetry profiles
 - Nucleation density; growth rate; morphology of the product; etc.



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Nucleation-and-Growth Mechanism

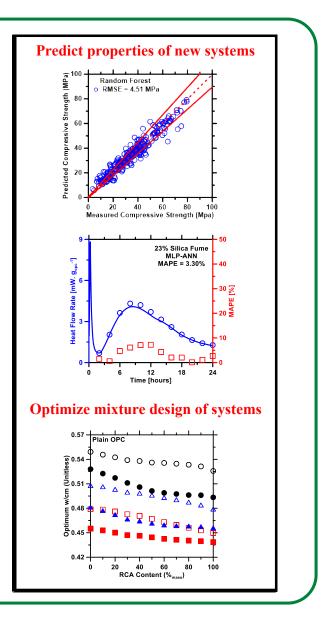
- Modified phase boundary nucleation-and-growth models
 - Product morphology is acicular
 - Product does not grow freely into capillary pore space
 - Density of product not constant
 - Product growth rate varies with time (or supersaturation)
- Excellent fits can be obtained
- But, calorimetry profiles still needed to optimize parameters





Supervised machine learning

- Machine learning (ML) is a form of artificial intelligence: unsupervised and supervised
- **Supervised ML model** is first trained using a database
- The ML model develops patterns, input-output correlations in the data
 - Correlations may or may not be known from theory
- Once trained, the ML model can leverage inputoutput correlations to predict in new datadomains
- In the case of cement pastes/concretes
 - Training: ML model learns correlations between inputs (physiochemical properties of precursors) and output (heat evolution; elastic modulus; strength; etc.)
 - Testing: Predicts properties (heat evolution) of new pastes/concretes, using their mix design as input





Database used for machine learning

- Training: hundreds of unique pastes
 - Plain, binary, and ternary pastes
- Cement: Commercial and synthetic
- SCMs
 - Different types: limestone; quartz; calcined clay; silica fume; fly ash; etc.
 - 0-to-60%_{mass} cement replacement
- Inputs: Physiochemical properties (composition, PSD) of pastes at t = 0 h
- Outputs: growth rate of hydrates
- Blind-testing: Pastes, with new PSDs and replacement levels of SCMs; different cement compositions

Input: PSDs, composition, and contents of cement and SCMs; & water content D e Pr 25 KG est Carcura ns Output: Growth rate of hydrates Hydrate's Outward Growth Rate - PC + 18%SF **?** 1.6 PC + 35%LS + 23%MK pBNG Model Output: Heat Flow Rate Simula ຜັ 0.8 б 0.4 0.0 18 24 6 12 Time (hour)



Advanced Nucleation-and-Growth Model

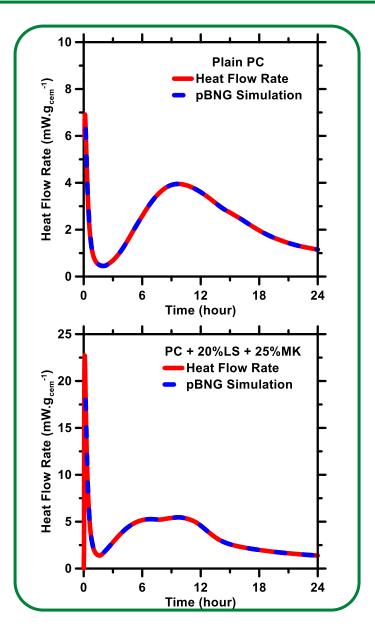
• $\alpha(t) = B \cdot X(t)$ Degree of hydration obtained from calorimetry profiles

•
$$X(t) = 1 - \exp\left(-2r_G \cdot G_{out}(t) \cdot a_{BV} \cdot t \cdot (f_{Out}(t) \cdot a_{BV} \cdot t)\right)$$

$$\left(1 - \frac{F_D(G_{out}(t) \cdot \sqrt{\pi \cdot g \cdot I_{density} \cdot t})}{G_{out}(t) \cdot \sqrt{\pi \cdot g \cdot I_{density} \cdot t}}\right)$$
Calculate the growth rate of hydrates using Cahn's eugation

•
$$SSA_{binder} = SSA_{cement} + a_{scm1}SSA_{scm1} \frac{z_{scm1}}{100 - z_{scm1}} + a_{scm1}SSA_{scm2} \frac{z_{scm2}}{100 - z_{scm2}}$$

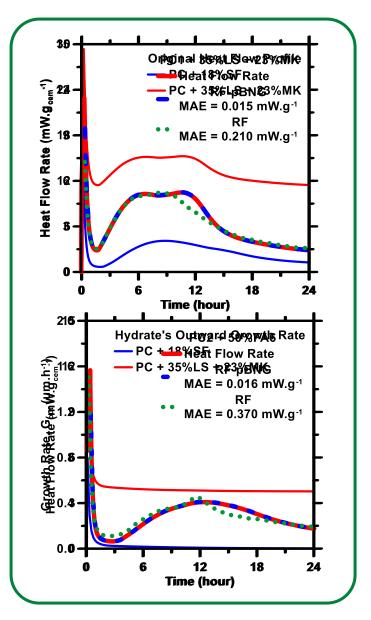
- Effective surface area (a_{scm}) is used to account for pozzolanic reaction, sulfate effect, Al(OH)⁻, amalgamation and filler effect.
- This model can be used to simulate hydration kinetics of cement replaced by common SCMs





Advanced Nucleation-and-Growth Model

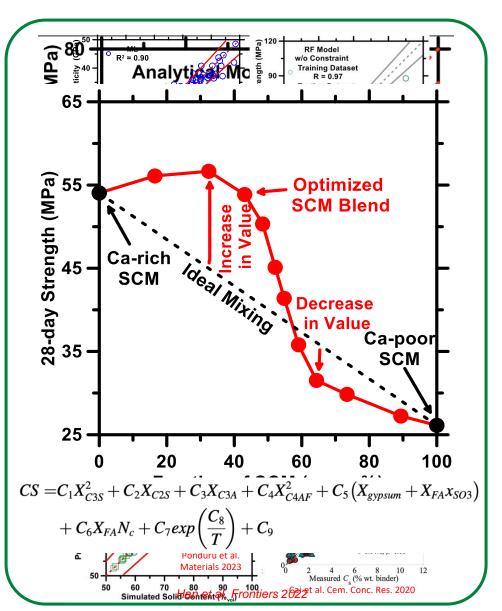
- Phase boundary nucleation and growth models require hydrate's growth rate (G_{out}) to reproduce heat flow profiles
- G_{out} profile showed simpler structure and trend
- ML optimized parameters for pBNG model, which allowed it to reproduce the heat flow rate profiles
- ML integrated with pBNG model produced superior predictions to standalone ML
- Nucleation and growth model regulate final output to avoid violation of material laws.





Other applications of machine learning

- Combine with *game theory* to develop simple, closed-form analytical models
 - (1) Quantify "importance" of each input parameter
 - (2) Construct a function using only "consequential" parameters
 - (3) Optimize the coefficients
- Predict constructability and compliance metrics of concrete
 - set time; rheology; steel rebar corrosion potential; strength
- Material design to optimize performance



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Conclusions

- Using SCMs to partially replace cement become an emerging solution for reducing the carbon emission in the cement industry
- Nucleation and growth model is the promising tool to reproduce hydration kinetics of [cement + SCMs] systems
- It cannot produce hydration kinetics for new systems without knowing growth rate of hydrates
- Harness the power of machine learning to predict necessary parameters for the nucleation and growth model
- The advanced nucleation and growth model can predict the hydration kinetics of [cement + SCMs] systems without the violation of material laws



Questions?

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