

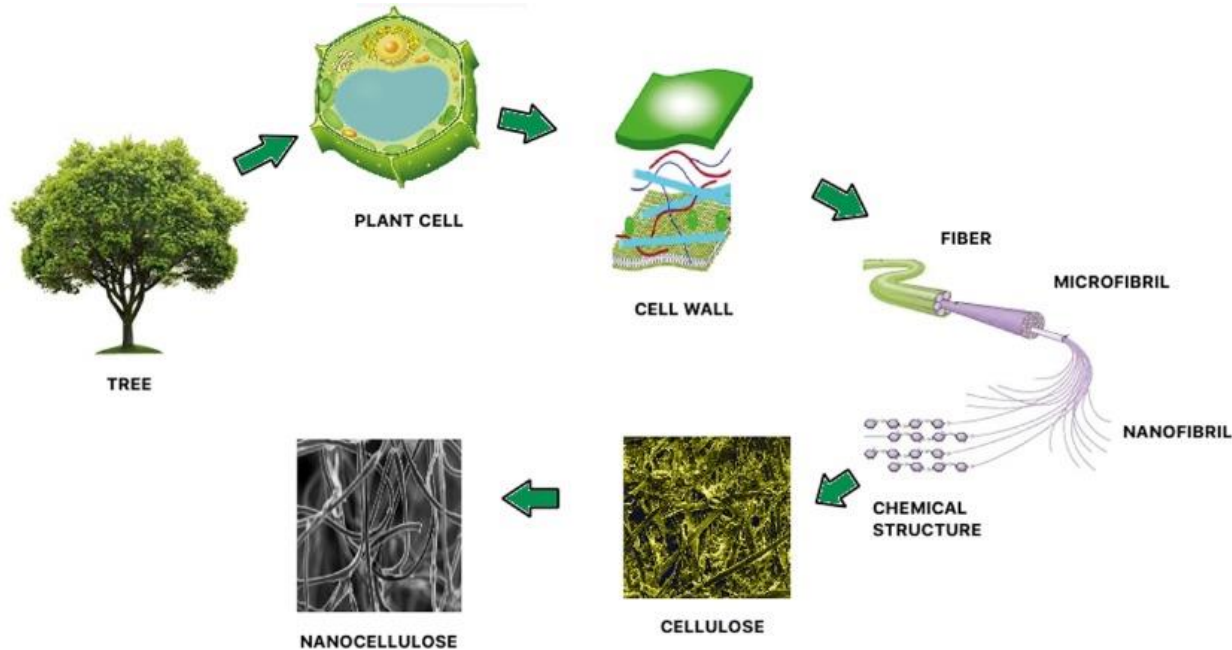
Application of Bio Based Nano Fibers as Mechanical and Durability Performance Enhancements for Cement Based Composites

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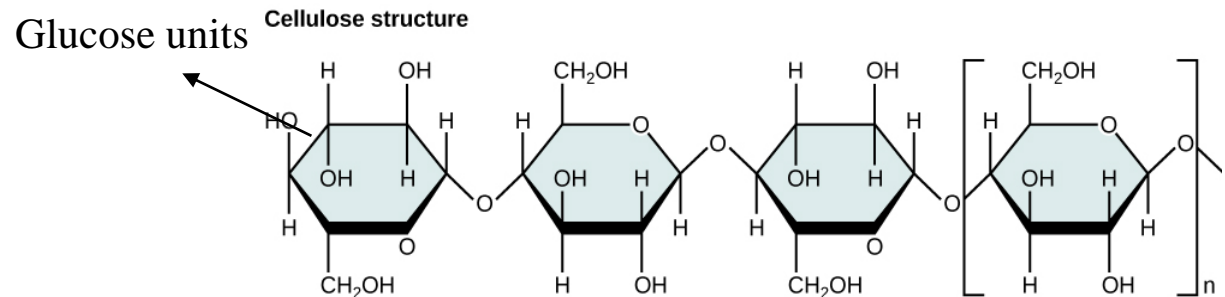
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❖ Cellulose nanofibrils (CNF)

- Most abundant biopolymer
- High tensile strength
- Reactive surface for functionalization
- 10 to 50 nm diameter, length can be up to a few micrometers
- Hydrophilic nature
- Tunable properties



<https://doi.org/10.1016/B978-0-12-823963-6.00012-0>



Research Objectives

1. Effects of CNF on Cement hydration
2. Interaction between cement pore solution and CNFs
3. Effects of CNF on sulfate resistance
4. Effects of CNF on alkali-silica reaction (ASR)

Experimental Setup

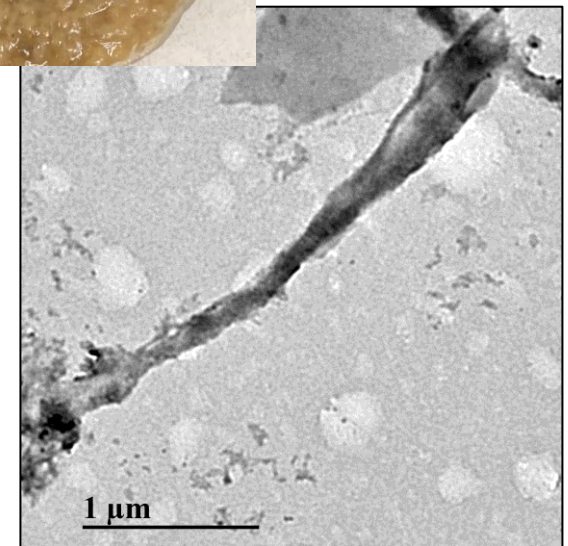
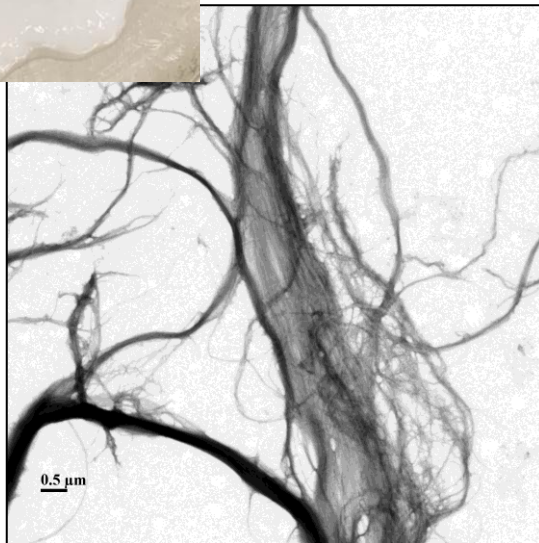
- Fibril Diameter: 80-300 nm
- Fibril Length: 100-500 μm
- Aspect Ratio: 800-1000 L/D
- DCNF was 34% solid in liquid form and LCNF was 29% solid in liquid form.



Cellulose type	Cellulose Doses (% of cement weight)	Binder
Control	0%	Type I cement
L-CNF	0.05%	
	0.1%	
	0.3%	
D-CNF	0.05%	
	0.1%	
	0.3%	

- ❑ **LCNF (Lignocellulose nanofibrils)** – made from unbleached (contains lignin) wood without any pre-treatment
- ❑ **DCNF (Delignified cellulose nanofibrils)** – made from bleached (i.e., delignified) wood pulp

Cellulose Nanofibers



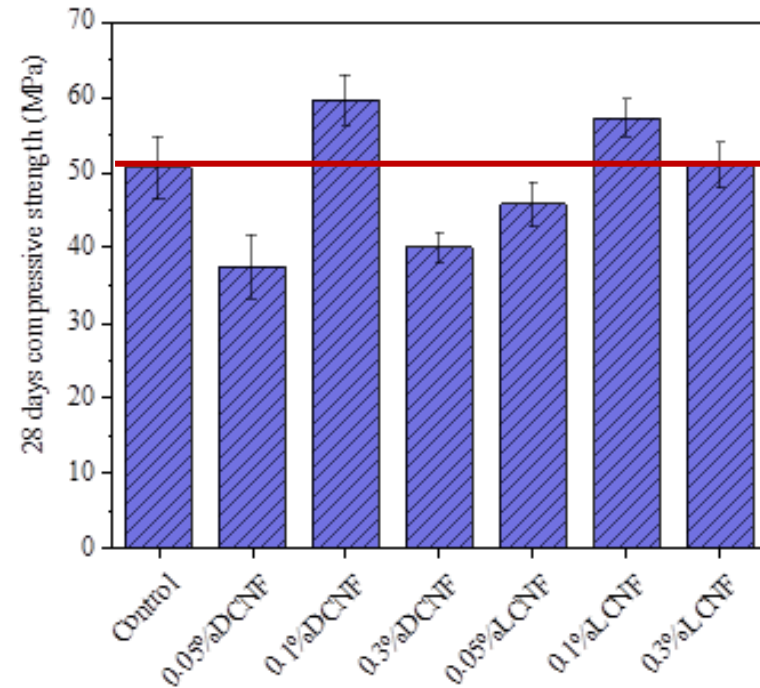
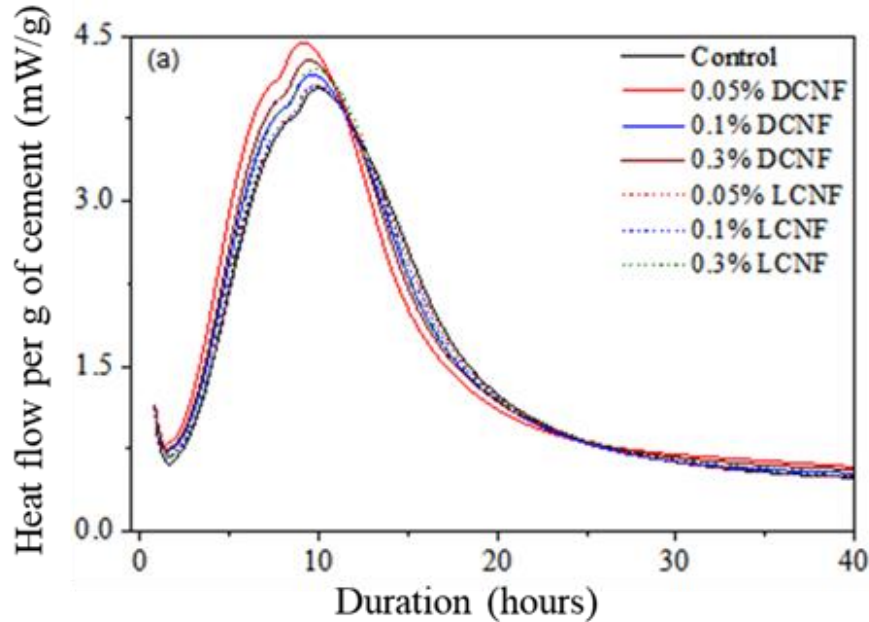
Cellulose nanofibers from bleached wood pulp (DCNF)

Cellulose nanofibers from unbleached wood pulp (LCNF)



Effects of CNF on Cement Hydration and Compressive Strengths

Before Exposure Parameters



Effects of CNF on hydration

Effects of CNF on strength

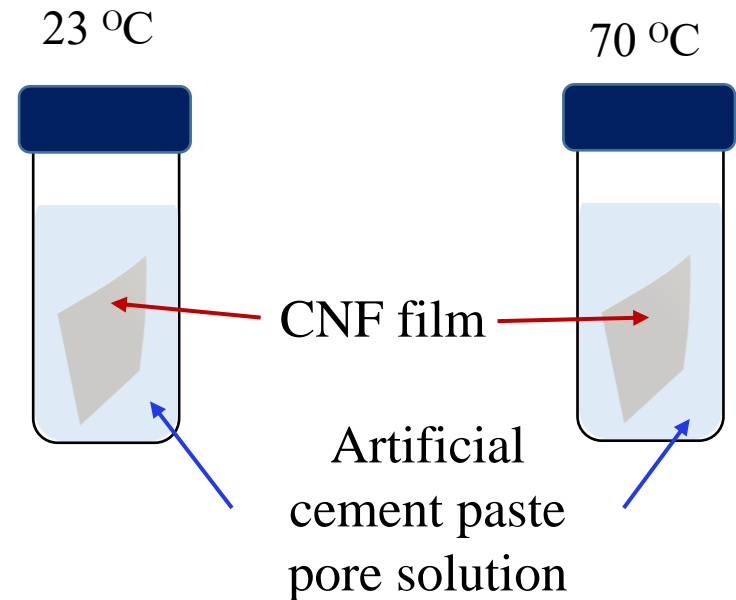
Takeaway:

- (i) Addition of CNF does not significantly affect the early age hydration reaction
- (ii) 0.1% CNF dosages slightly improved 28-days compressive strength



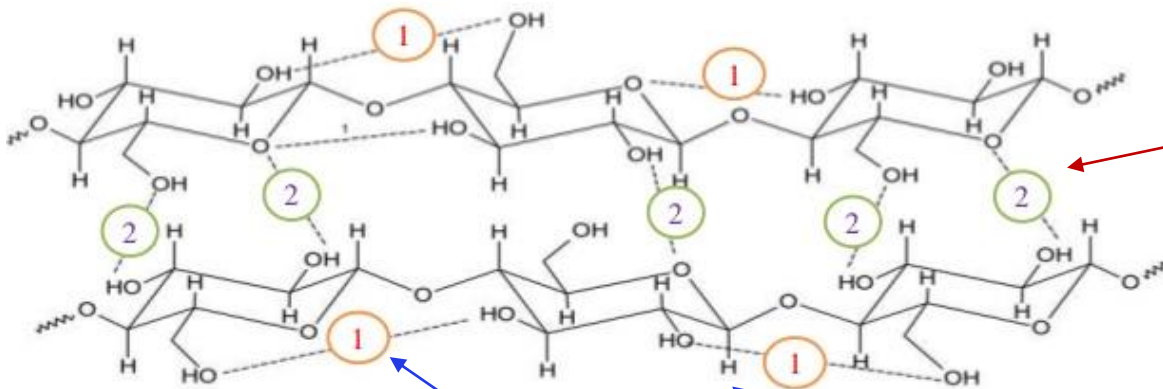
Interactions between Cement Pore Solution and CNFs

- ❑ 28 days artificial pore solution (APS) of cement
- ❑ Chemical composition of APS: $\text{Ca}(\text{OH})_2$, Na_2SO_4 , NaCl , NaOH , KOH , Al_2O_3
- ❑ Curing conditions :
 - ❑ Regular curing : 23°C
 - ❑ Temperature curing : 70°C



- ❑ Pore solution chemistry was monitored using (Inductively coupled plasma - optical emission spectrometry (ICP OES))
- ❑ Cellulose film was analyzed using Fourier-transform infrared spectroscopy (FTIR)

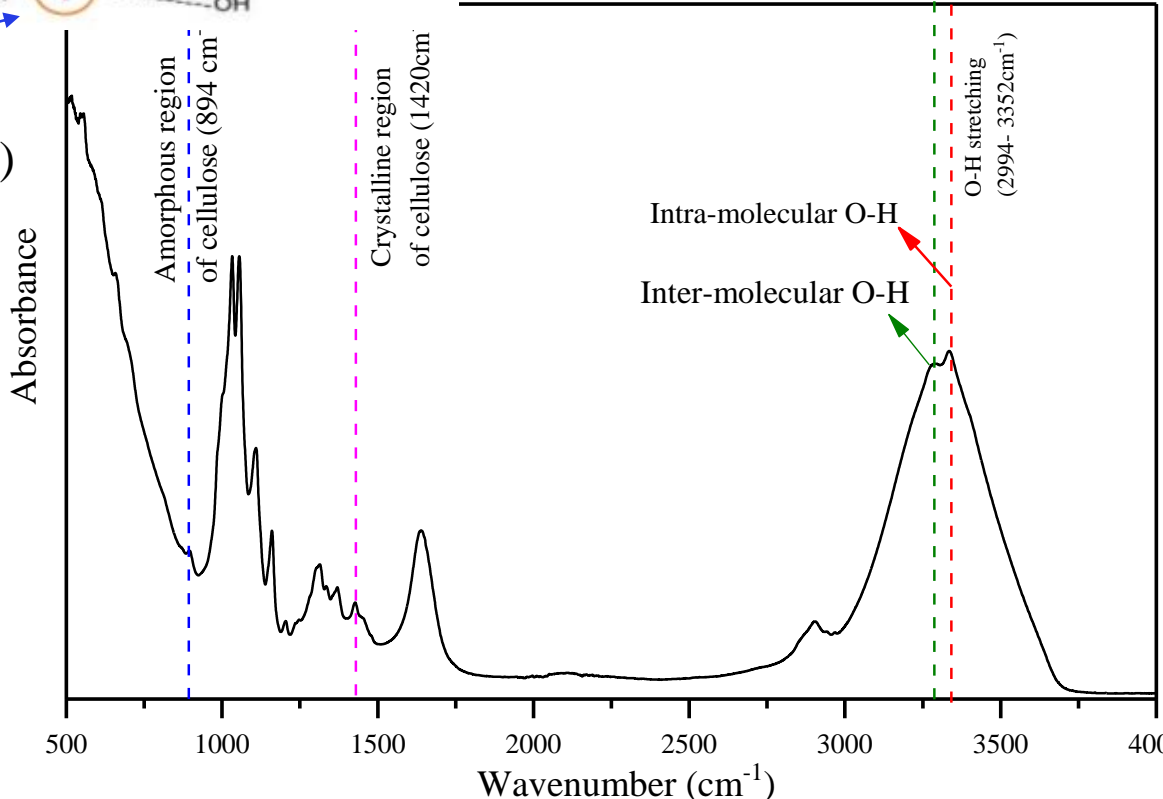
Effects of Pore Solution on CNF

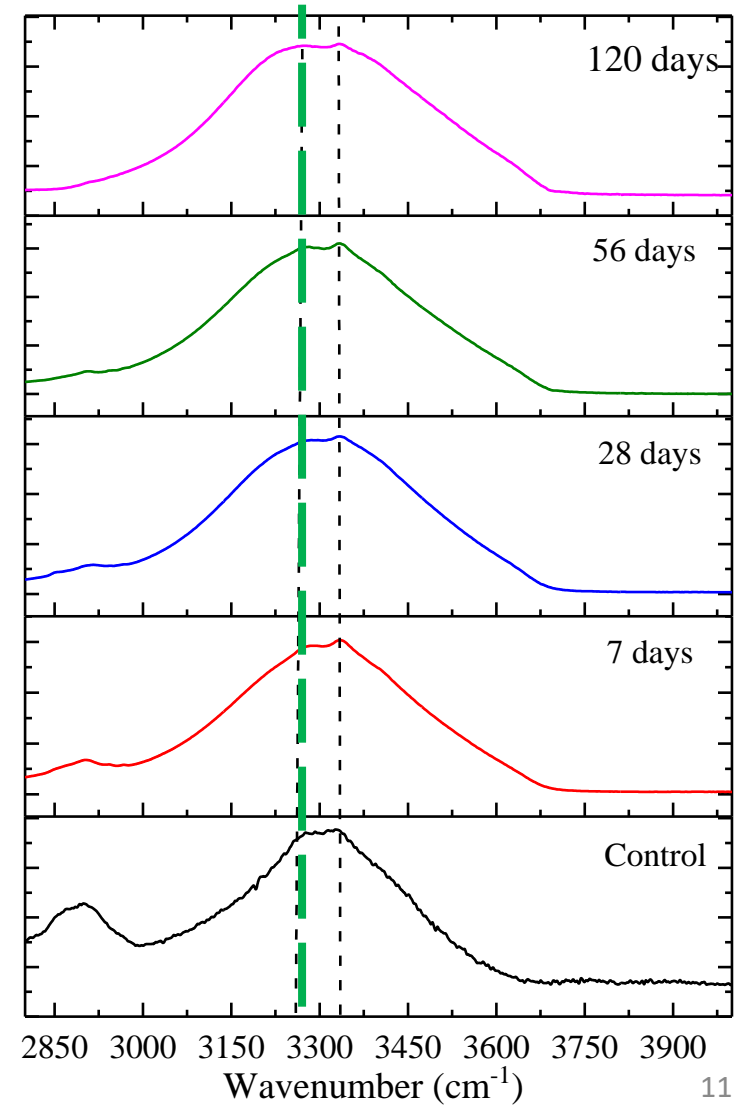
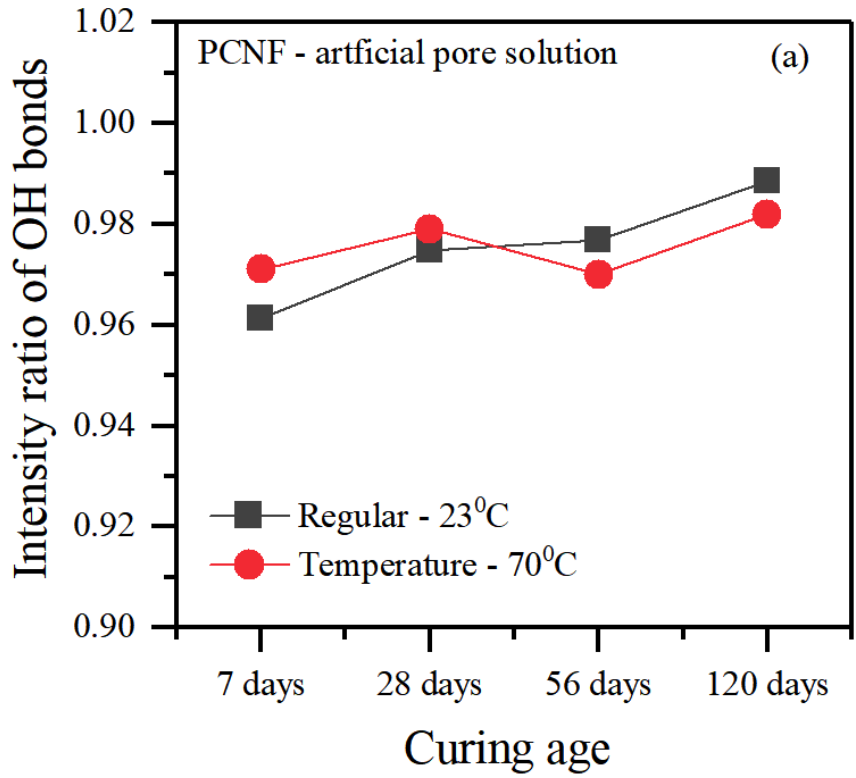


Intra OH Bond
(within a cellulose chain)

Inter OH Bond
(between two cellulose chains)

➤ Increase in intra OH bond indicates breaking of cellulose chain



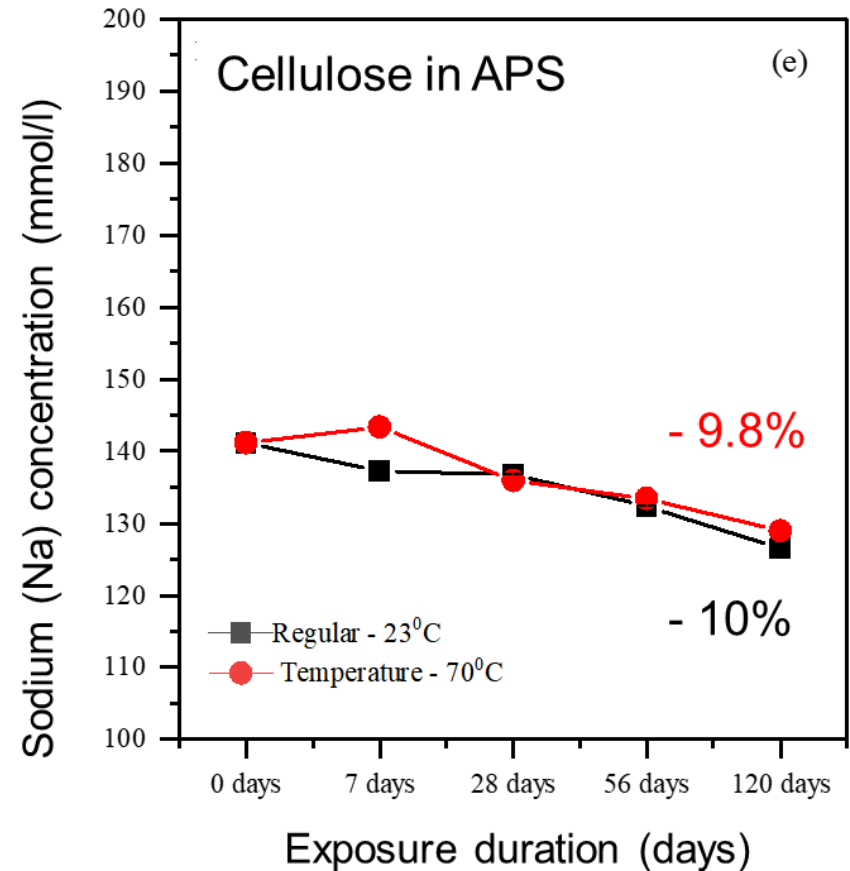
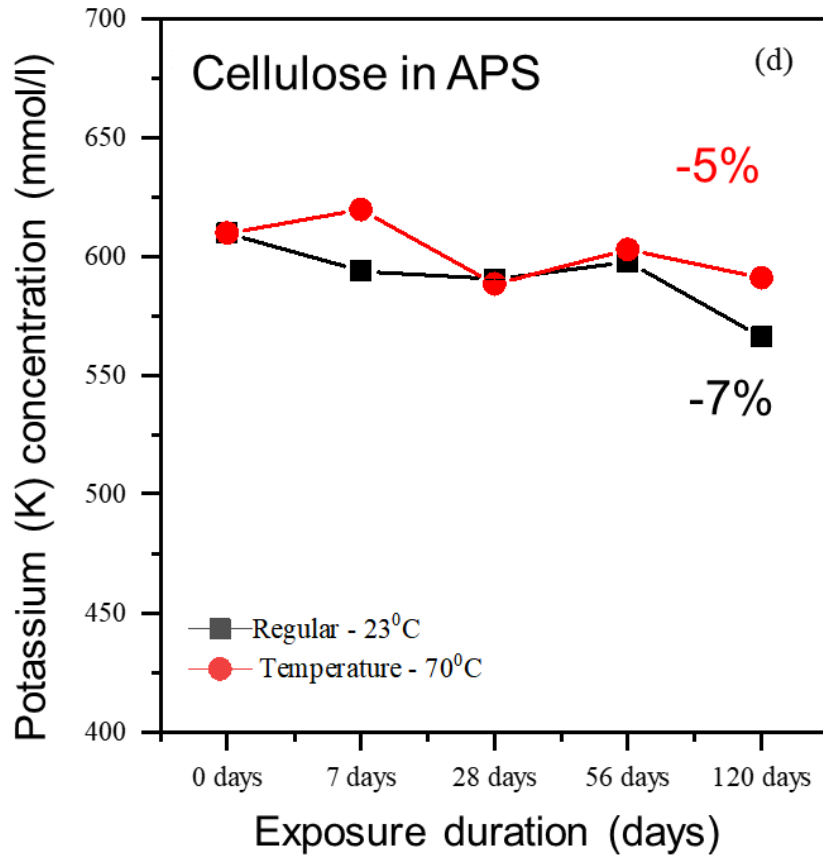


$$\text{Intensity ratio} = \frac{\text{Intensity of inter O-H}}{\text{Intensity of intra O-H}}$$

➤ Transformation of cellulose from amorphous phase → crystalline phase



Effects of CNF on pore solution



- No other secondary phase precipitation was observed using X-ray diffraction



Resistance to Sulfate Attack

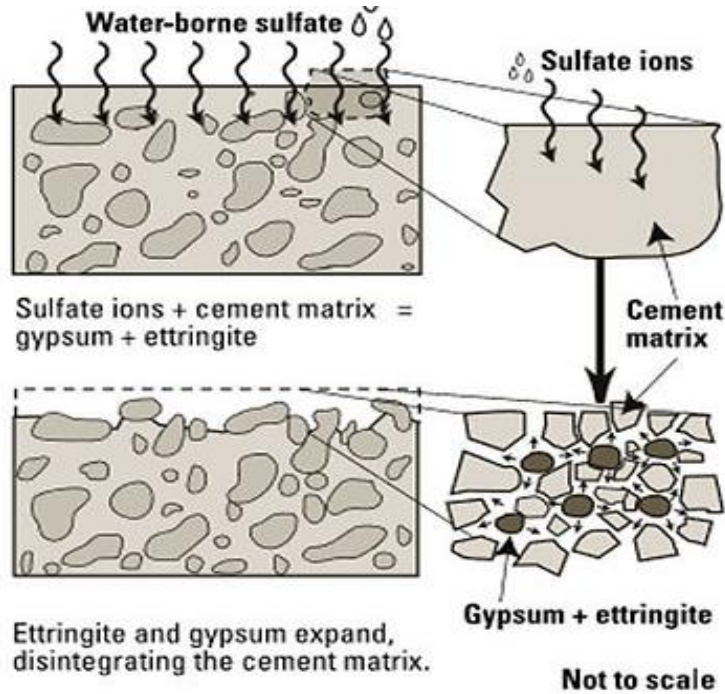
Sulfate attack in concrete:

Cause and Significance



Mechanism

www.ctlgrouppqatar.com



Appearance



Concrete exposed to sulfate rich soil



Concrete exposed to sulfate rich water

- **Current Practice:** Use of specialty cement (i.e., Type V cement)

Experimental plan for sulfate attack



Length change (ASTM C 1012 standard)

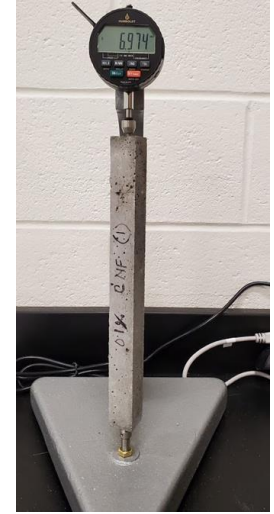
5% Na_2SO_4

Strength variation

Compressive strength on paste samples (28 days, 28+15, 28+1month, 28+3months, 28+6 months)

Microstructure

TGA, XRD on paste samples (28 days, 28+15, 28+1month, 28+3months, 28+6 months)



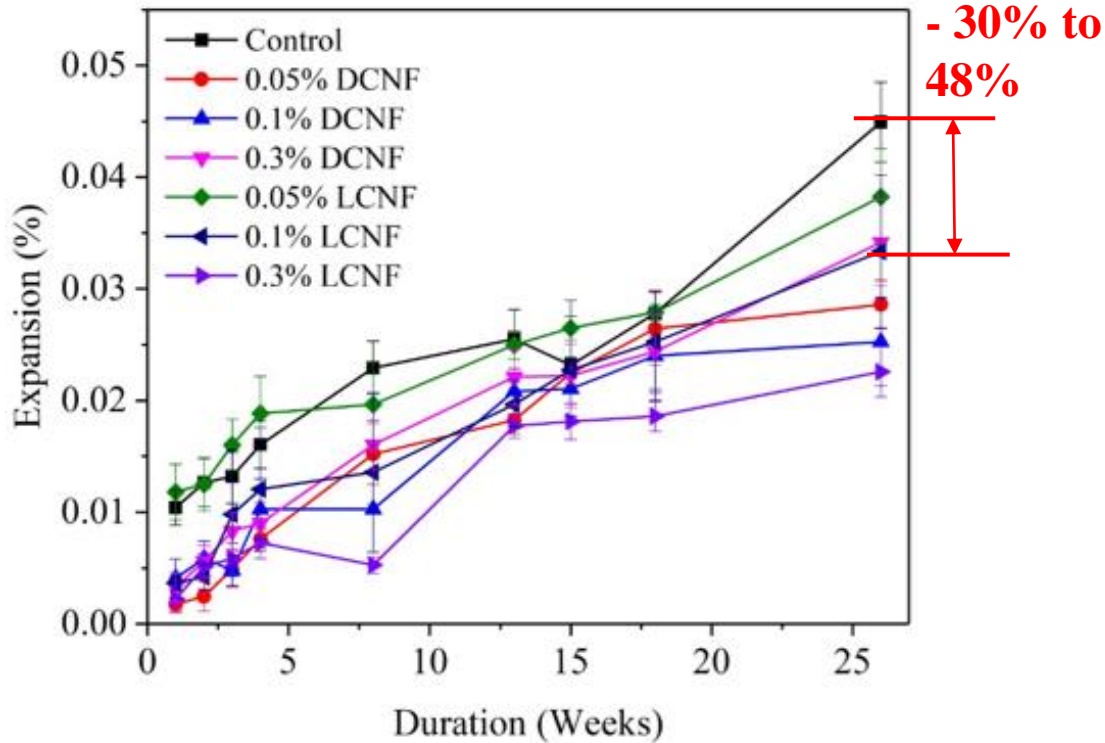
Mortar bars for expansion measurement



Mortar cubes for strength measurement

Samples exposed to Na_2SO_4 solution

Length Change

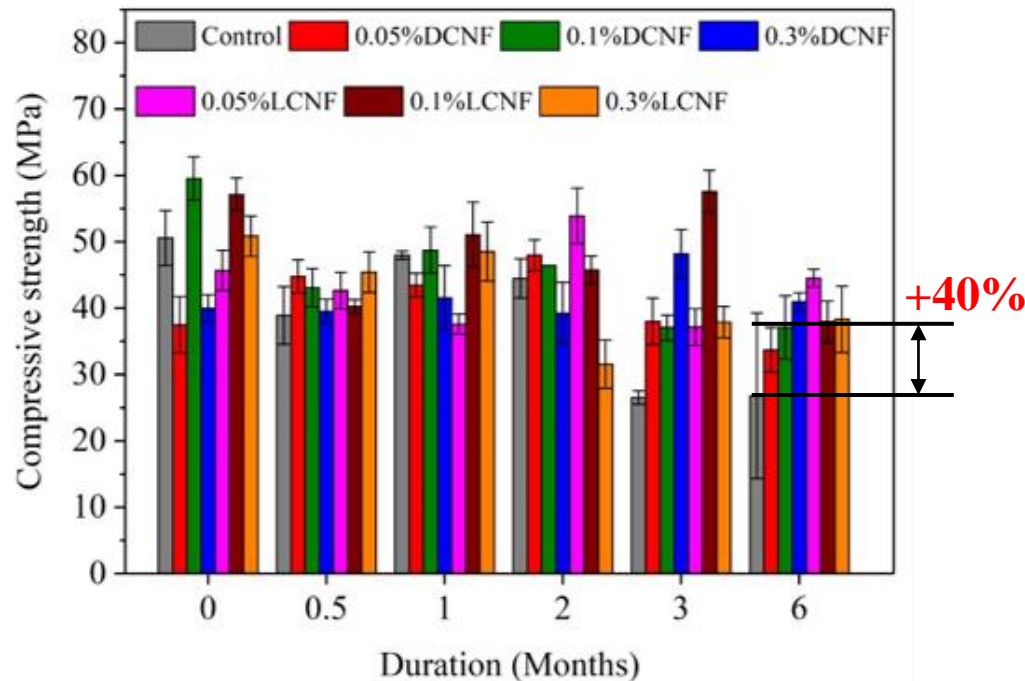


Sample ID	Expansion reduction (%) w.r.t control batch after 6 months in Na_2SO_4
DCNF 0.05%	46.67
DCNF 0.10%	47.88
DCNF 0.30%	18.38
LCNF 0.05%	25.25
LCNF 0.10%	33.73
LCNF 0.30%	50.9

Takeaway: Sample expansion was reduced by 30 to 50% due to the addition 0.1% CNF

Samples exposed to Na_2SO_4 solution

Compressive strength



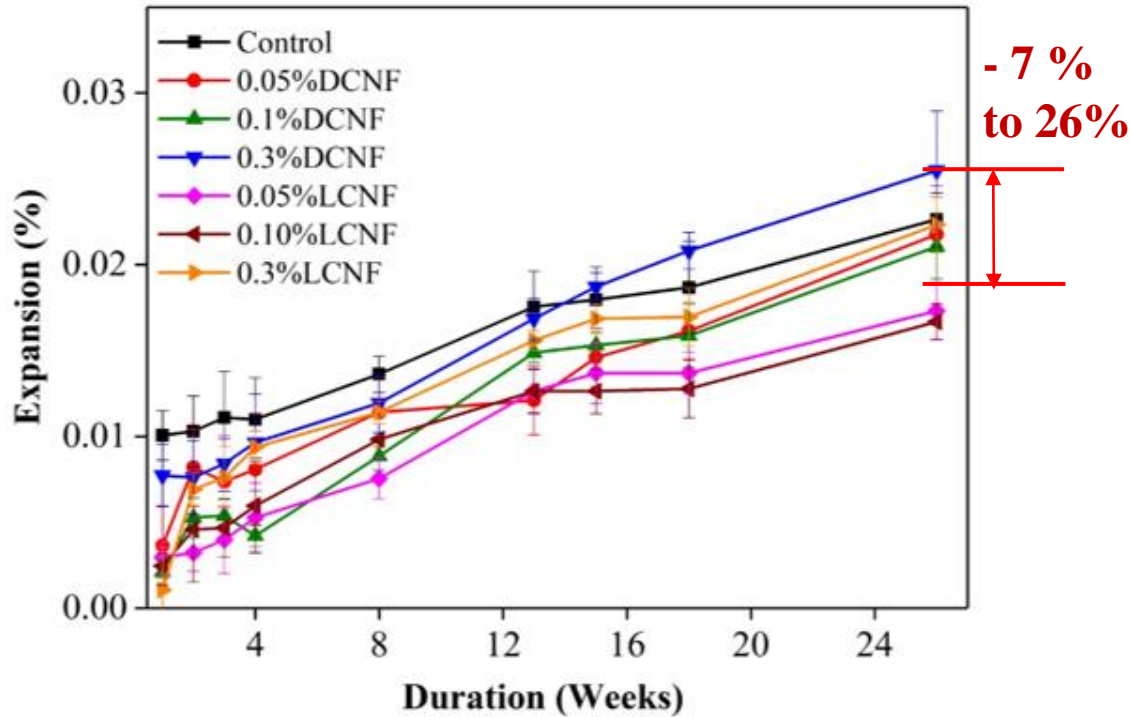
Sample ID	Change in compressive strength after 6 months exposure	
	w.r.t. control	w.r.t. before exposure strength
Control	----	- 47.01
DCNF 0.05%	+ 25.68	- 10.11
DCNF 0.10%	+ 38.39	- 37.66
DCNF 0.30%	+ 52.83	+ 2.58
LCNF 0.05%	+ 66.08	+ 2.55
LCNF 0.10%	+ 41.46	- 33.62
LCNF 0.30%	+ 42.91	- 24.65

Takeaway:

- (i) There was no specific trend with dosage
- (i) After 6 months exposure, control batch (without CNF) had the lowest strength

Samples exposed to $MgSO_4$ solution

Length Change



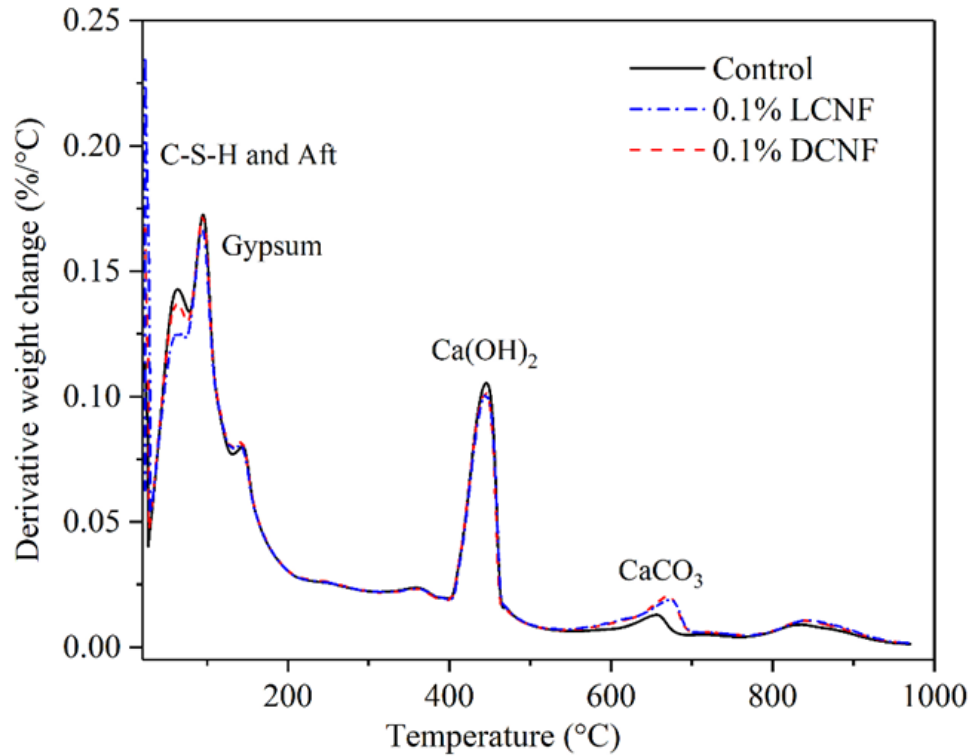
Sample ID	Expansion reduction (%) w.r.t control batch after 6 months in Na_2SO_4
DCNF 0.05%	-3.87
DCNF 0.10%	-6.97
DCNF 0.30%	+ 12.66
LCNF 0.05%	-23.51
LCNF 0.10%	-26.36
LCNF 0.30%	-1.39

Takeaway: Sample expansion was reduced by 7 to 26% due to the addition 0.1% CNF

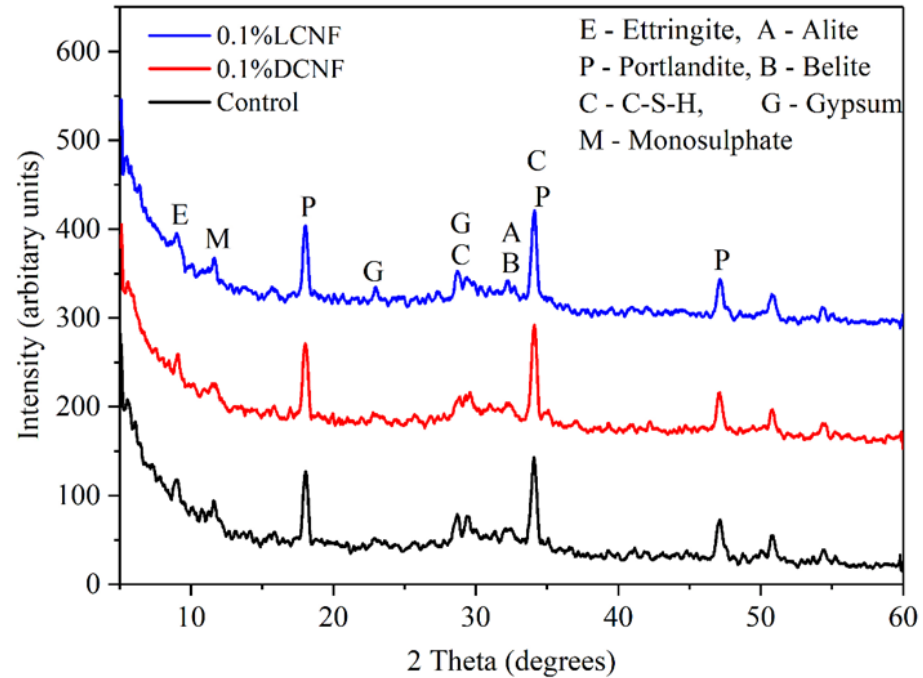


Microstructure

Thermogravimetric analysis (TGA)



X-ray Diffraction (XRD)

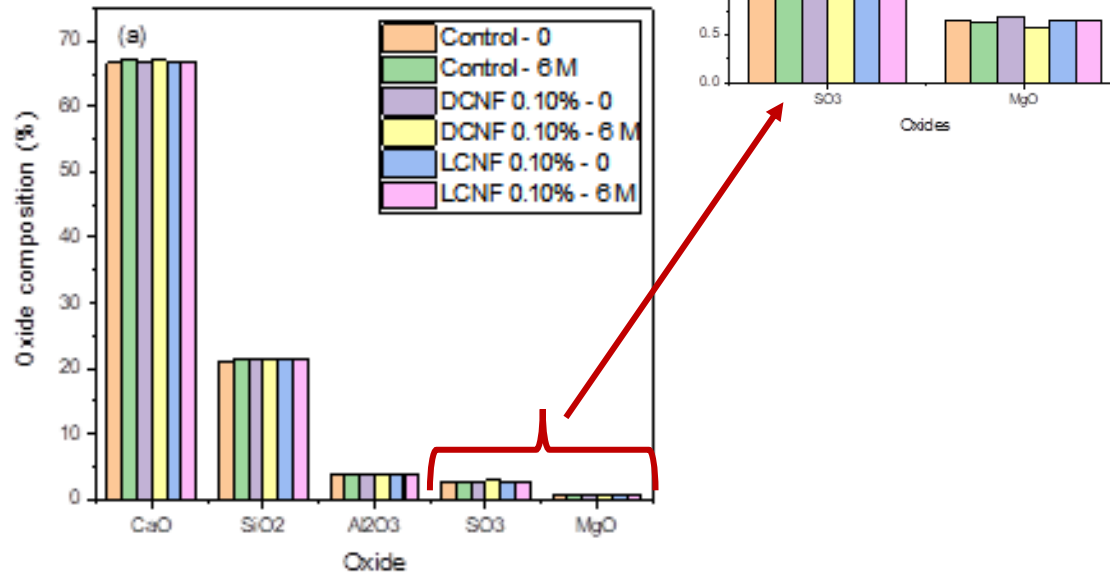


Takeaway:

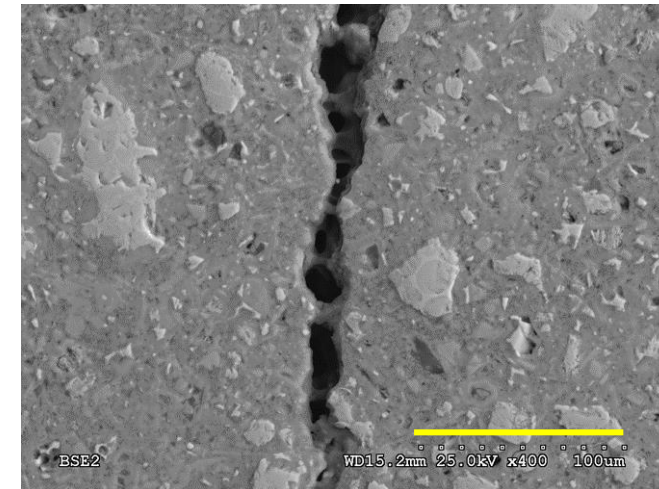
The addition of CNF reduced gypsum and ettringite formations.

Microstructure

X-Ray Fluorescence (XRF)



Scanning electron microscopy (SEM)



Takeaway: The addition of CNF improves resistance to sulfate damage by (i) reducing gypsum and ettringite formations, and (ii) bridging the cracks.

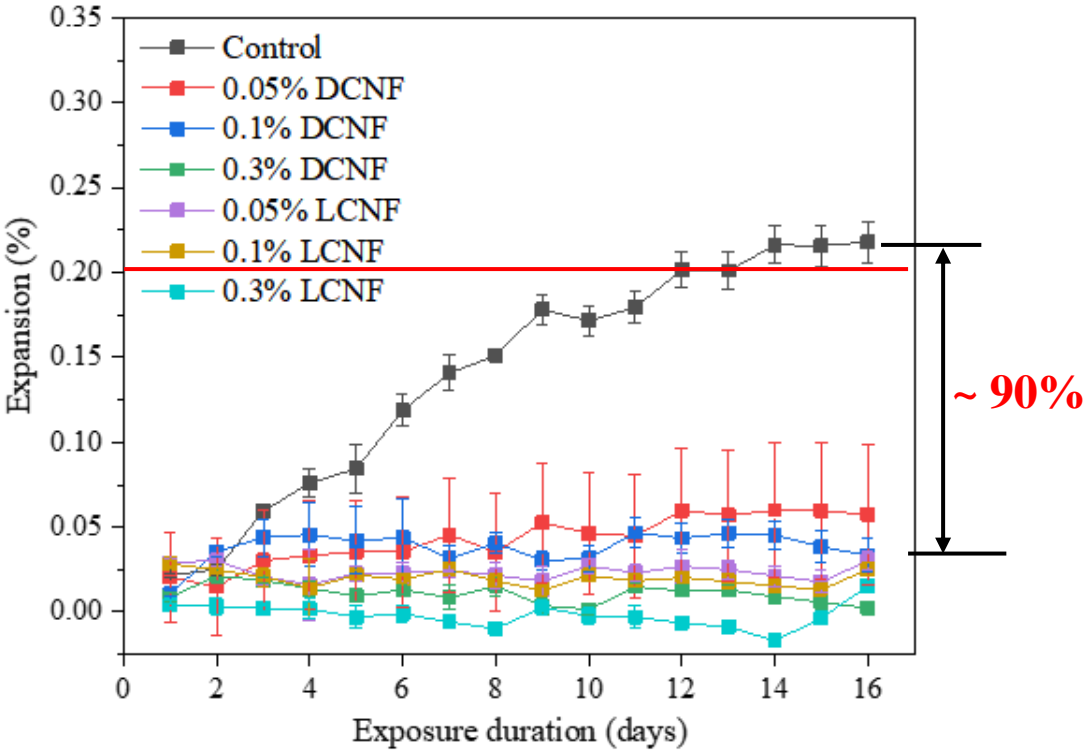
Resistance to Alkali Silica Reaction (ASR)

ASR Susceptibility test

- ❑ Test method: ASTM C 1260
- ❑ Water to cement ratio = 0.5
- ❑ Aggregate type: sodium borosilicate
- ❑ Bar immersed in 0.1N NaOH solution at 80°C
- ❑ Measurements taken at 24 hours interval up to 16 days



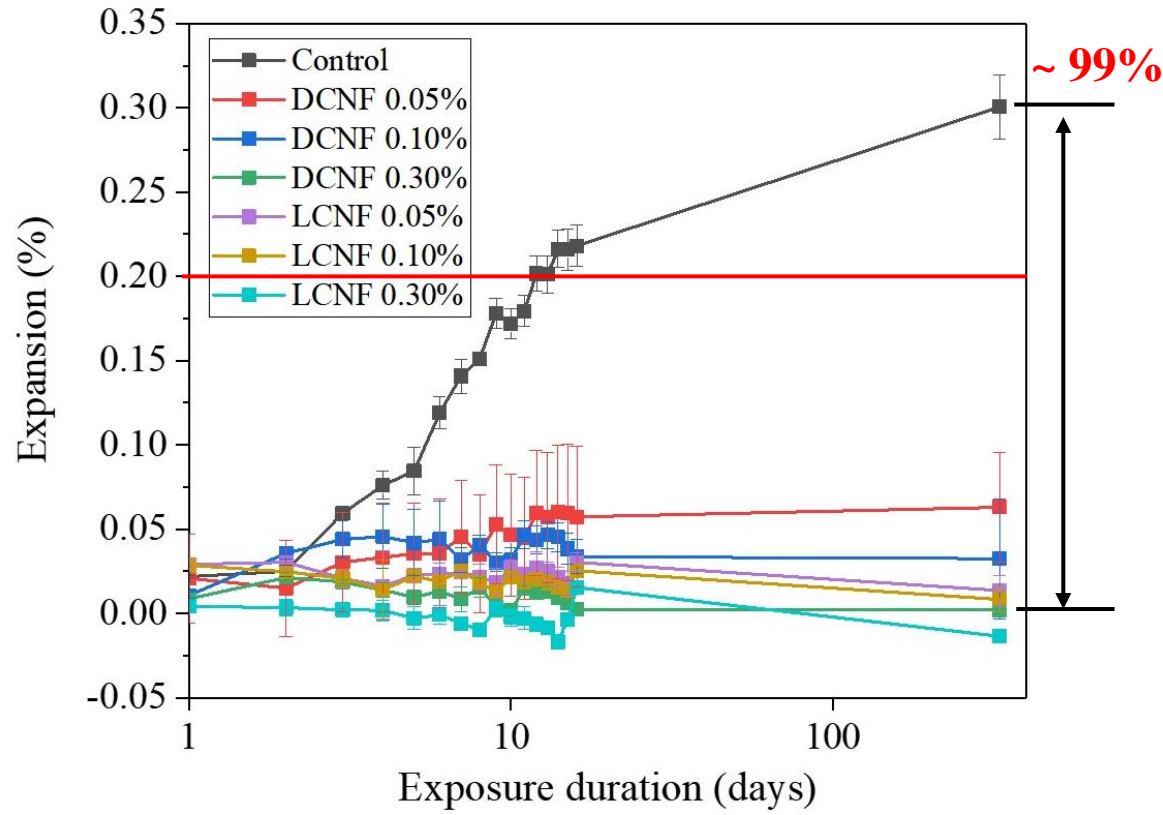
Effects of CNF on Alkali-Silica Reaction



Sample ID	Expansion reduction (%) w.r.t control batch after ASR reaction
DCNF 0.05%	73.71
DCNF 0.10%	84.59
DCNF 0.30%	98.89
LCNF 0.05%	86.06
LCNF 0.10%	88.38
LCNF 0.30%	92.90

Takeaway: Sample expansion was reduced by 80 to 90% due to the addition 0.1% CNF

Effects of CNF on Alkali-Silica Reaction

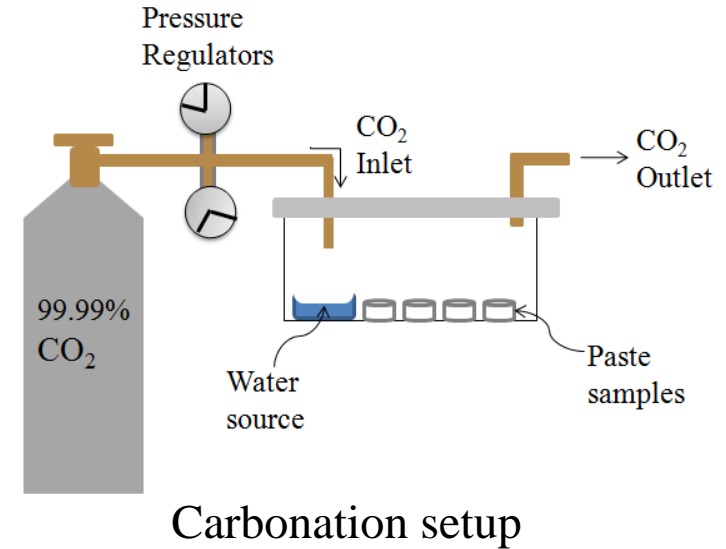
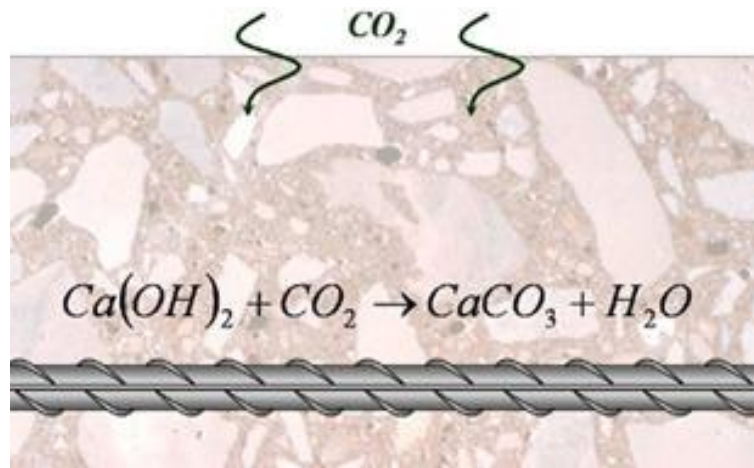


Sample ID	Expansion reduction (%) w.r.t control batch after ASR reaction
DCNF 0.05%	79.0
DCNF 0.10%	89.2
DCNF 0.30%	99.3
LCNF 0.05%	95.5
LCNF 0.10%	97.2
LCNF 0.30%	104.5

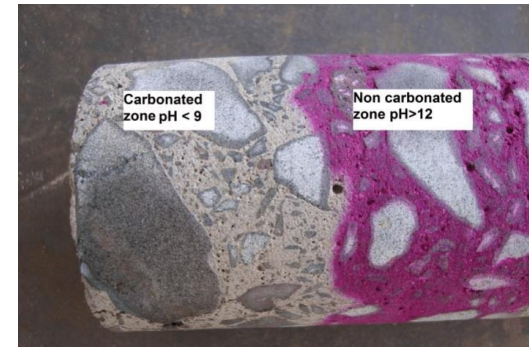
Takeaway: Sample expansion was reduced by 99% due to the addition of 0.1% CNF

Resistance to Atmospheric Carbonation

Atmospheric Carbonation: Cause and Significance

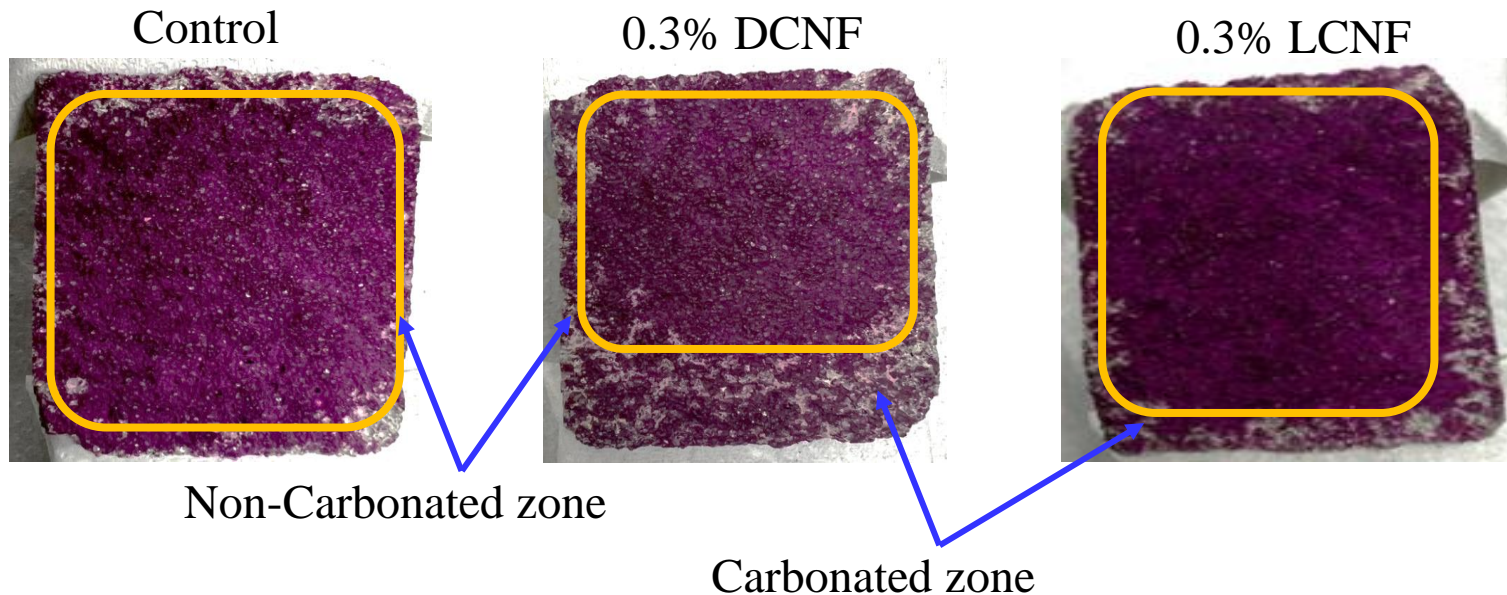


The carbonation reaction reduces the pH of the pore solution to less than 9, which leads to the destruction of the passive oxide layer protecting the steel bars from corrosion



Phenolphthalein indicator test

28 days carbonation



Takeaway: (i) Effects of CNF on atmospheric carbonation resistance of mortar samples were inconclusive. (ii) For certain cases, the addition of CNF appeared to increase the carbonation.

Conclusions

- ❑ The addition of CNF significantly (~ 50%) reduced sulfate damage in mortars.
- ❑ For most of the durability tests, LCNF containing samples performed slightly better (~ 10%) than the DCNF
- ❑ Due to the alkaline hydrolysis, the degree of crystallinity of CNF increases with time when exposed artificial pore solution
- ❑ The addition of CNF reduced the expansion of mortar samples due to the ASR by more than 90%.
- ❑ For durability performance enhancement, 0.1% dosage of CNF was found to be optimum.
- ❑ Additional studies are required to understand the effects of CNF on carbonation. This can be helpful in increasing carbon sequestration in concrete.

P³Nano



Thank you!

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