

Performance of concrete containing water-hyacinth ash (WHA) as cement replacement – resistance to elevated temperature and seawater exposures

Ahmed Omran

Nancy Soliman

Mohamed Mahgoub

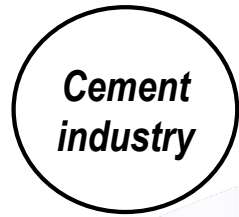


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Outline

- Introduction
- WHA
 - Production
 - Properties
 - Strength-activity index (SAI)
- Concrete mix design
- Compressive and tensile strengths
- **Effect of high temperatures on cube F_{cu} of WHA concrete**
- **Resistance of WHA concrete to seawater**
- Conclusions

1- Introduction



5-8% of global CO₂ emission



Coal-based power plants are being shut down all over the world



Alternative Supplementary Cementitious Materials (ASCM)
By-products, waste, bio-based



2- WH plants

Origin :

- Lives in fresh water (rivers) in warm regions
- Native to Latin America, Introduced to USA in 19th (Louisiana), then moved to central states

Problematic :

- Fast growing rate (2 plants produce 30 plants in 23 days and 1200 plants in 4 months)
- Block water ways → increase water level in main & branch drainages and flood agricultural land
- Damage canal walls
- Increase water evaporation
- Decrease oxygen in water → 1-acre WH = 40 persons
- Enhance mosquito production

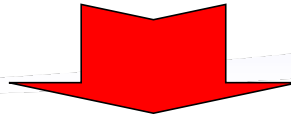


WH

- Growth & decomposition of WH cause **organic pollution and costly \$\$\$**
- Collecting & burning in open air → **air pollution**

Cement

- Shortage in:
 - resources of raw materials
 - energy needed for manufacture
- CO₂ emission



What about using WH as ASCM ?

WHA production



Sieving WHA on sieve # 200



Los - Angles (LA) apparatus for grinding



WH plants in fresh case



Collection of WH plants from watercourses

Furnace



Open-air burning

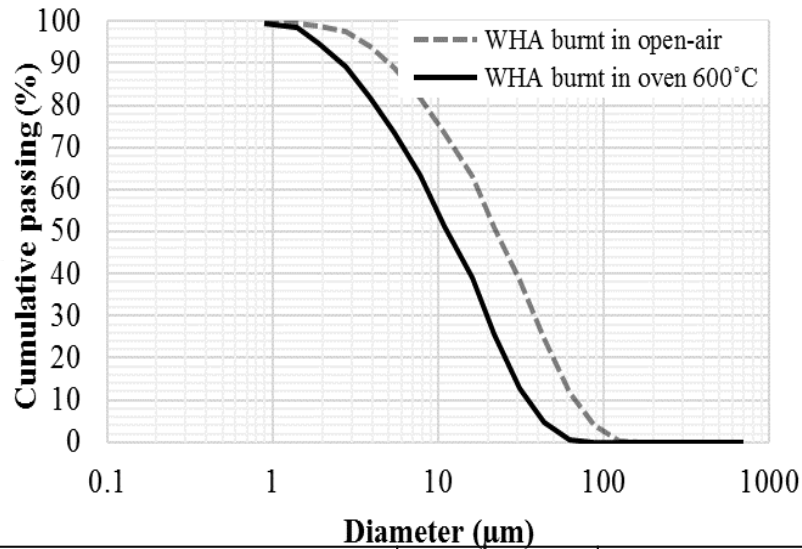


Oven burning at 600°C [1112°F]



WH plants after drying for 2-3 weeks

Physio-chemical analysis of WHA



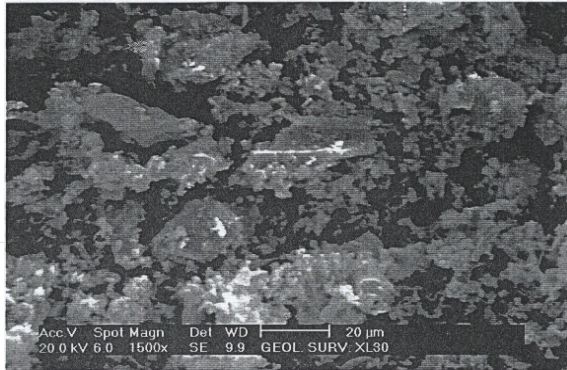
Σ 3 oxides ≈ 50% (as fly and rice-husk ash)

Constituent	%	
	Open-air burning	Oven burning at 600°C [1112°F]
SiO₂	33.89	34.50
Ti ₂ O ₃	0.75	0.78
Al₂O₃	6.77	6.95
Fe₂O₃	5.77	6.02
MgO	5.40	5.93
CaO	10.08	11.46
Na ₂ O	1.26	1.41
K ₂ O	9.83	10.98
MnO	0.66	0.73
P ₂ O ₅	1.04	1.13
Cl ⁻	3.82	4.02
SO ₄ ²⁻	2.37	3.74
LOI	17.93	11.91

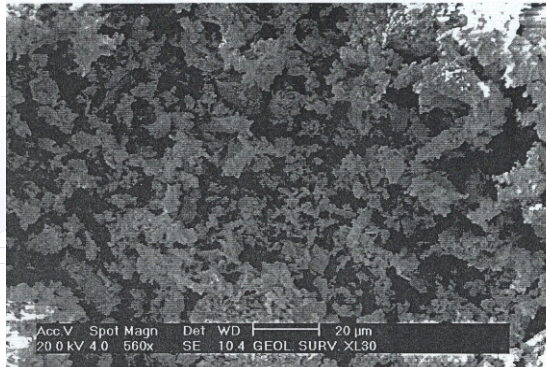
Large LOI due to organic materials

	Open-air burning	Oven burning at 600°C [1112° F]
% wt. of WHA/dried plant	27.65	25.85
D₅₀ (µm)	23	12
Specific gravity	2.52	2.65
Color	Dark gray	Light brown

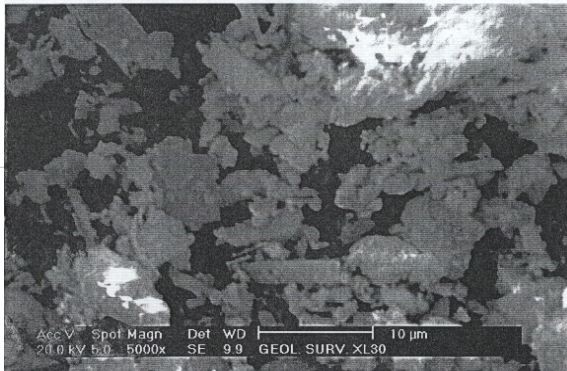
SEM of WHA



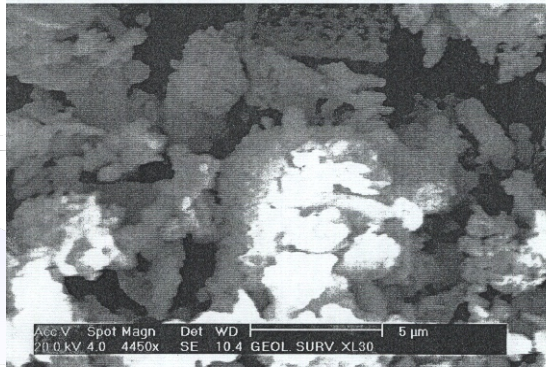
(a): At 1500 x



(a): At 560 x



(b): At 5000x



(b) At 4450 x

(A) WHA burnt in open air

(B) WHA burnt in oven 950°C

WHA is porous and mainly contain angular particles of irregular shape and rough textures, with few spherical particles of smooth surfaces

Strength-Activity Index (SAI) of WHA

Mortar mixtures

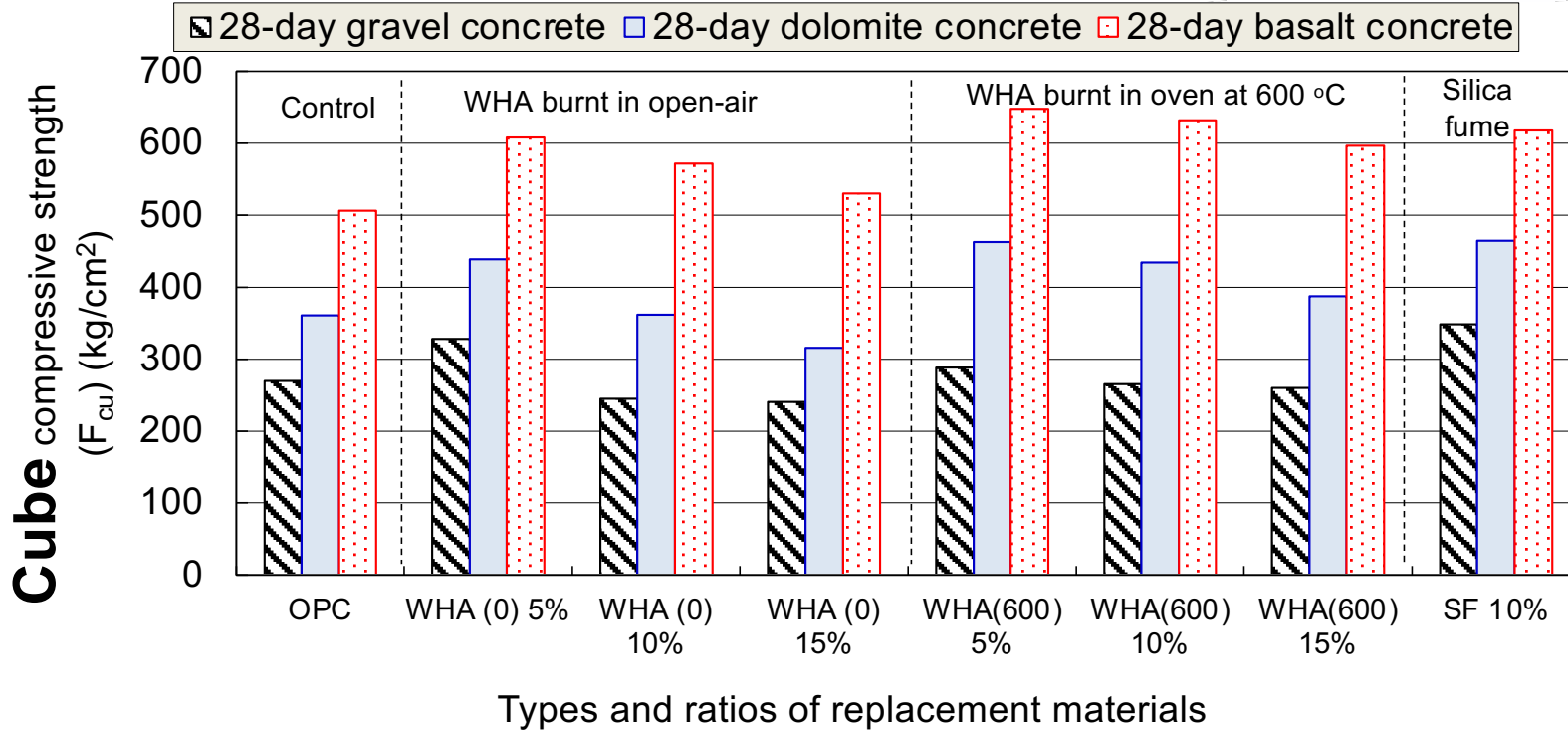
Materials	Replacement ratio	Pozzolanic activity index	
		7 days	28 days
Pure OPC	0%	1.00	1.00
WHA burnt in open air, $WHA_{(0)}$	5%	1.26	1.18
	10%	1.31	1.15
	15%	1.26	1.10
WHA burnt in oven at 600°C, $WHA_{(600)}$	5%	1.42	1.25
	10%	1.16	0.87
	15%	0.68	0.60
S.F.	10%	1.34	1.18

According to **ASTM C 618**, $SAI \geq 0.75 \rightarrow$ pozzolanic materials

3- Concrete mix design

Mix	Mix proportions (kg/m ³)							Rep/Cement
	Cement	Water	sand	Coarse aggregates 1- Gravel 2- Dolomite 3- Basalt	Replacement (Rep)			
					WHA ₍₀₎	WHA ₍₆₀₀₎	Silica fume	
OPC	300	150	638	1276	—	—	—	—
WHA ₍₀₎ 5%	285	150	637	1274	15	—	—	5%
WHA ₍₀₎ 10%	270	150	636	1272	30	—	—	10%
WHA ₍₀₎ 15%	255	150	635	1270	45	—	—	15%
WHA ₍₆₀₀₎ 5%	285	150	637	1274	—	15	—	5%
WHA ₍₆₀₀₎ 10%	270	150	637	1273	—	30	—	10%
WHA ₍₆₀₀₎ 15%	255	150	635	1271	—	45	—	15%
SF 10%	270	150	635	1269	—	—	30	10%

4- Compressive strength

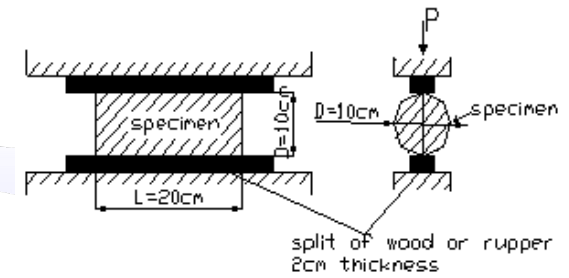
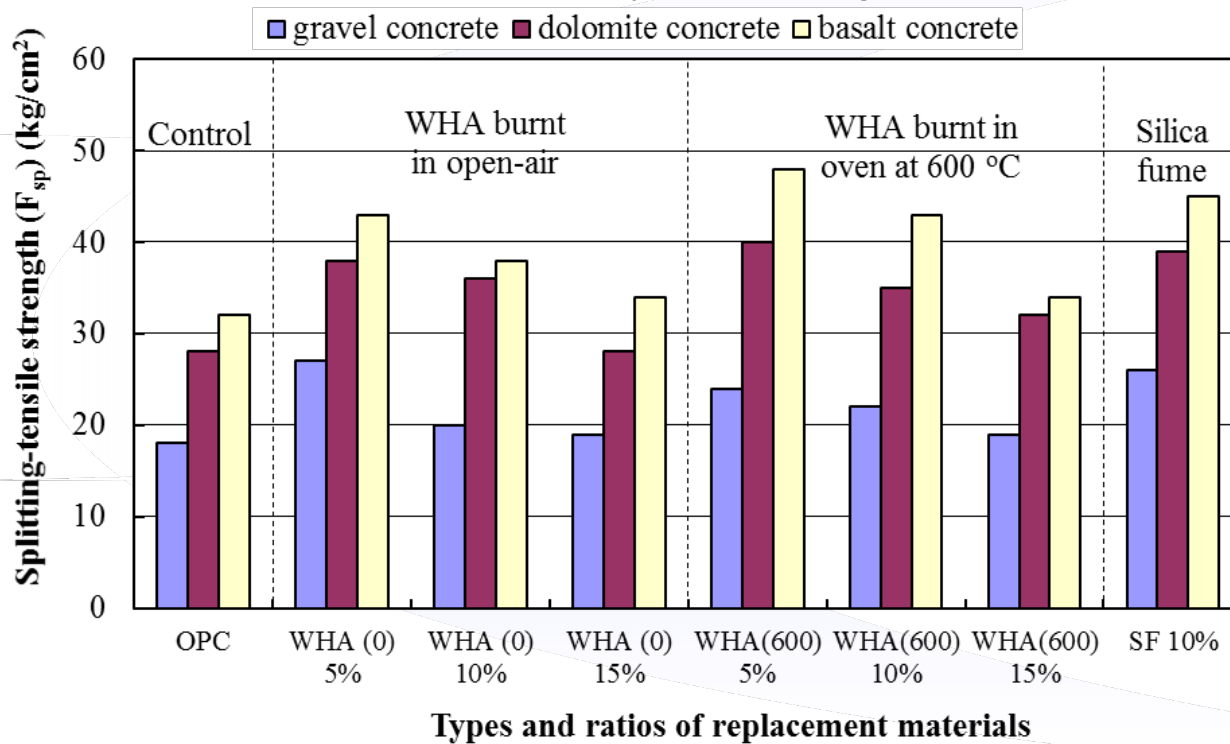


Cubes

Strength increase was due to pozzolanic reactivity & filling effect of WHA
 Burning method had no clear effect on strength

4- Splitting-tensile strength

After 28 days of curing



Cylinders

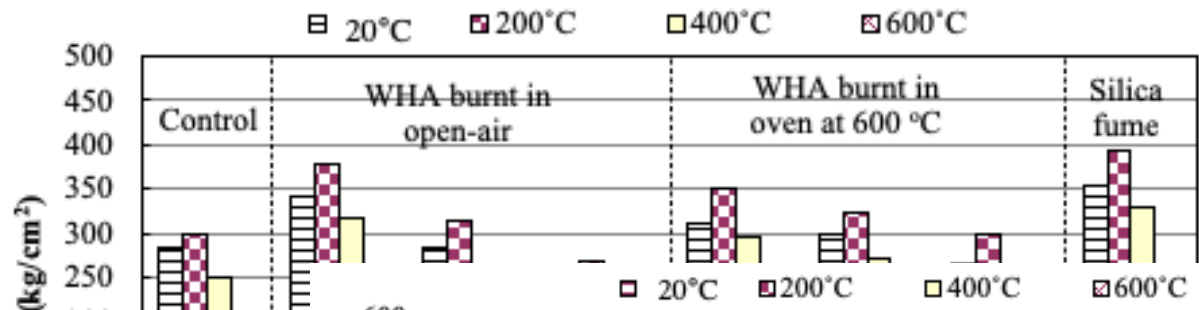
Pozzolanic reactivity, filling effect of WHA, and enhancing transition zone

5- Effect of high temperatures on cube F_{cu} of WHA concrete

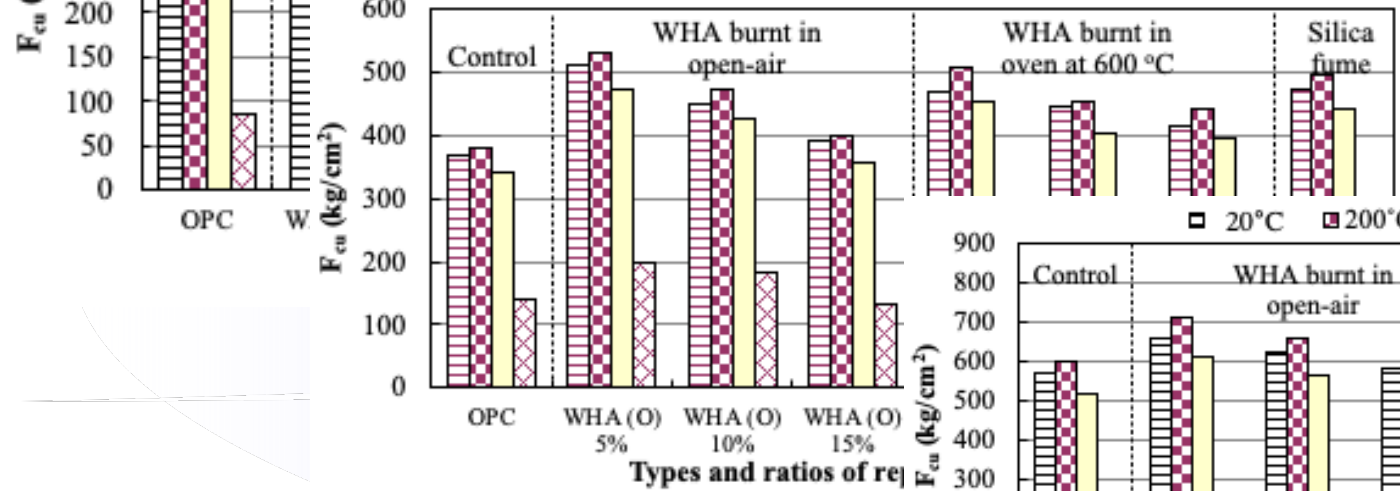
- 1-28 days, moisture curing (lime-water path at 23°C)
 - 28-91 days, lab atmosphere (23°C, 50% RH)
 - Oven drying at 105±5°C for 24 hrs
 - Furnace heating to 200, 400, 600°C (73, 392, 752, 1112°F) for **3 hrs**
 - Cooling to lab temperature before testing
- *Concrete at lab atmosphere (23°C, 50% RH) – served as a **reference***



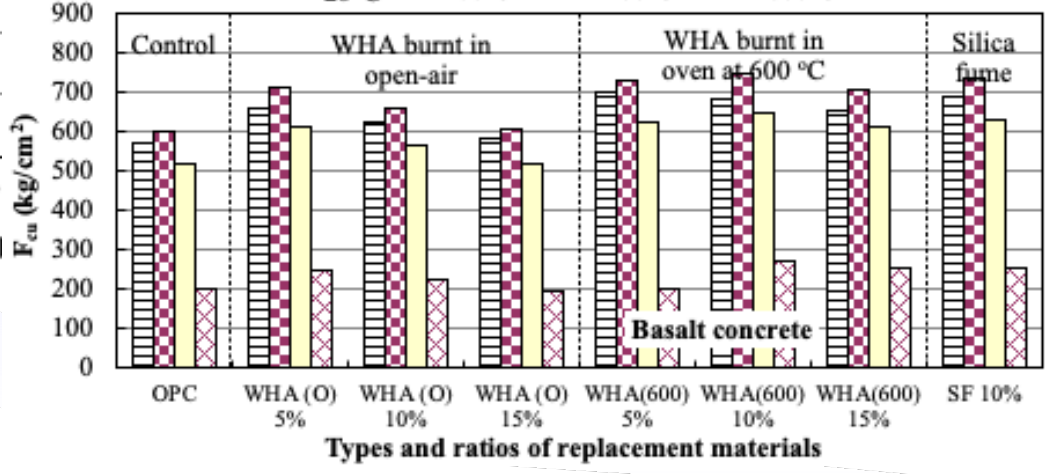
Furnace



Cubes



Types and ratios of replacement materials



Types and ratios of replacement materials

1- Cement paste unstable

Thermal treatments started at age of 3 months,

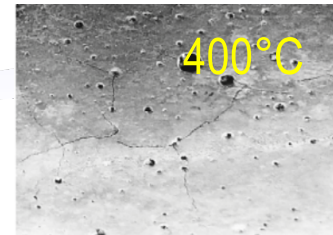
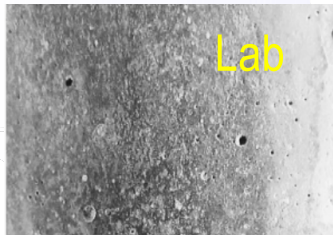
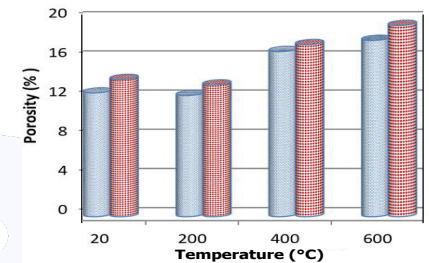
5-10%WHA was enough to react with ALL Ettringite, and convert it to CSH of higher thermal stability & binding forces

Extra WHA (10-15%) remains un-reacted (filler) as there is NO more Ettringite

100-300°C Heat causes internal moisture removal as faster rate, steam (so-called: internal autoclaving)

- internal pressure ↑
- porosity ↑
- transform any un-hydrated cement grains to Ettringite
weak if system contains only cement,
strong if pozzolanic material exists as transforms it to CSH

400-600°C, structure disruption and crack propagation



2- Aggregate more stable

Reduction in $F'_{cu} \propto$ thermal stability of aggregates (dolomite < gravel < dolomite)

6- Resistance of WHA concrete to seawater

Why seawater?

Sodium Na⁺



participate in alkali-aggregate reaction

12,000 mg/l

Chloride Cl⁻



promote steel corrosion

21,000 mg/l

Magnesium Mg⁺⁺



attack most constituents of hardened portland cement paste, especially aluminate constituent

1500 mg/l

Sulfate SO₄⁻

2600 mg/l

Curing Conditions:

1-28 days, moisture curing (lime-water path at 23°C)

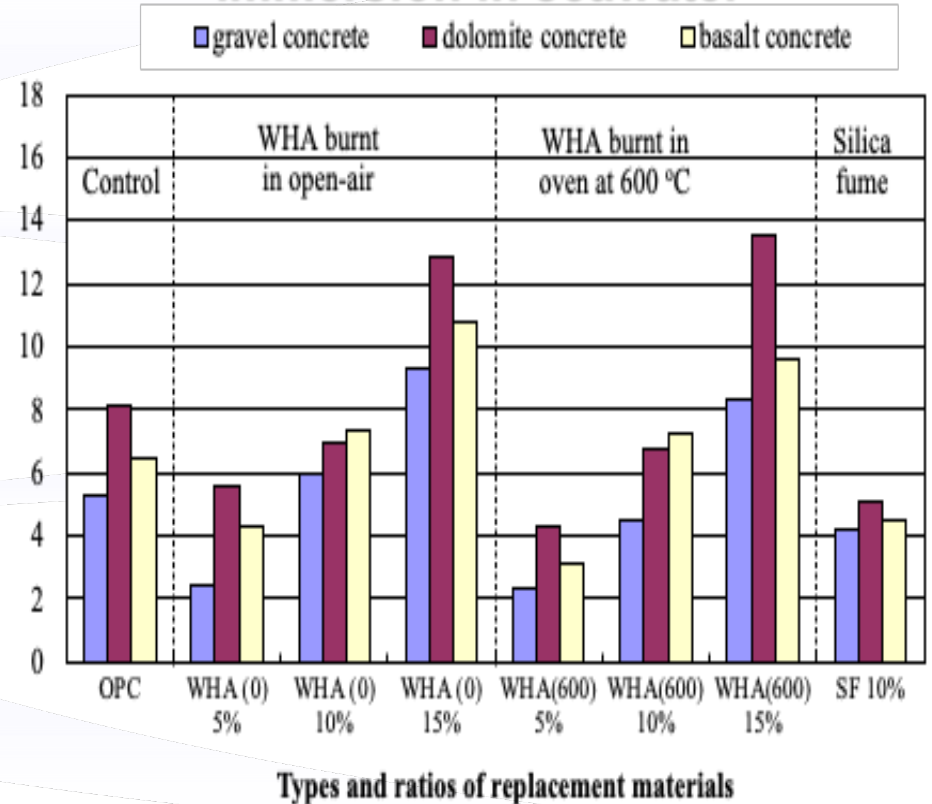
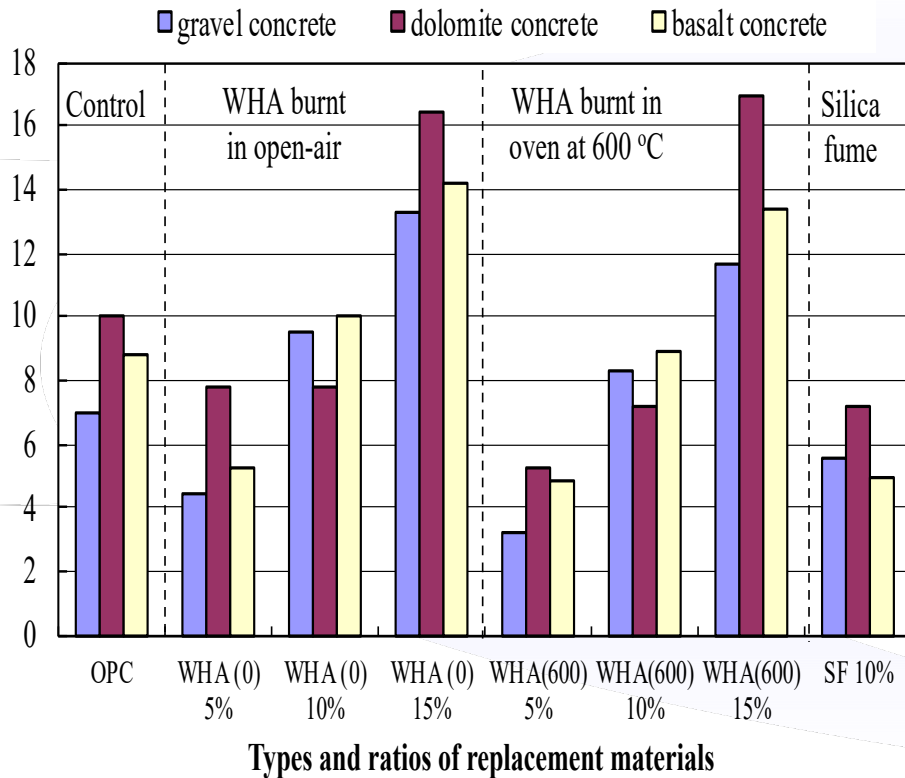
28-58 days, lab atmosphere (23°C, 50% RH)

32 days of special curing (age 58-90 days)

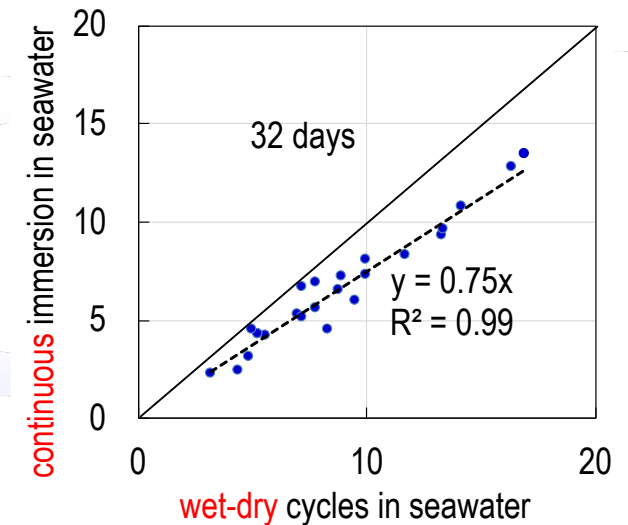
1. lab atmosphere (23°C, 50% RH) -- served as control
2. stored in seawater path for the entire period
3. dry-wet cycles: 1-day at lab atmosphere & 1-day in seawater path

% decrease in cube F'_{cu} (relative to lab atmosphere, control) - wet-dry cycles

% decrease in cube F'_{cu} (relative to lab atmosphere, control) - continuous immersion in seawater



- Wet-dry cycles more severe > continuous projection to seawater → due to change in volume and uniformity of water
- Densified microstructure (less porosity & permeability) of concrete containing WHA → prevent entry of ions
- Reduction of Ettringite (pozzolanic reactivity) that may deteriorate when reacts with SO_4^{--}
- Presence of more Al_2O_3 in WHA → it may act as expansive cement which recommended by ASTM C845 for marine structures (shrinkage compensation)



- Resistance of $\text{WHA}_{600} > \text{WHA}_0$ due to fineness
- Resistance of Gravel > Basalt > Dolomite, due to abrasion & surface texture

Conclusions

Using WHA as an ASCM in concrete mixes: adds a value to a waste material, contributes to green environment, cost saving of cement manufacture, and saves materials resources

Burning WH plants in closed ovens produces ash with no effect on environment and with high silica content than burning in open air

Concrete with 5-10% WHA has higher strength

- in normal curing conditions, subjected to elevated temperatures, or when contact chlorides and sulfates of seawater

Because adding WHA to ordinary concrete:

- reacts with ettringite $[Ca(OH)]$ from cement hydration and forms more C-S-H (pozzolanic reactivity) of stronger binding forces and higher thermal stability
- densifies microstructure and enhances transition zone
- lowers porosity and permeability

Thank you!

